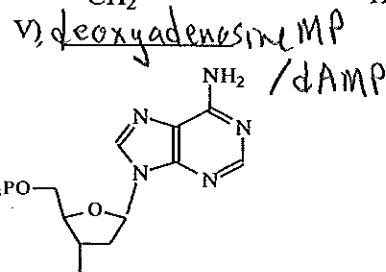
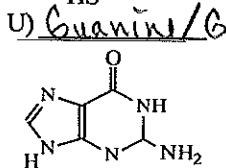
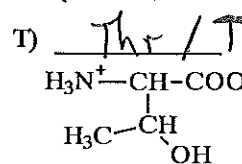
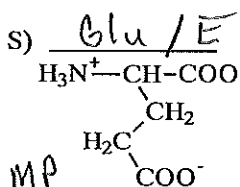
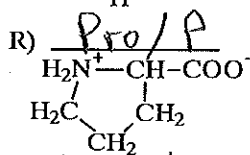
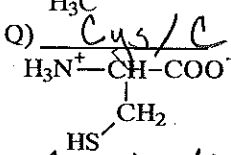
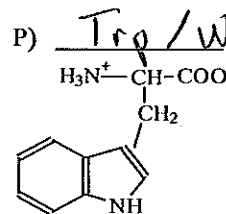
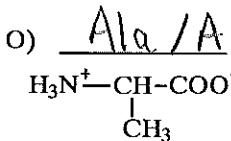
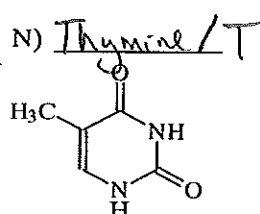
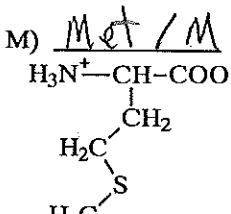
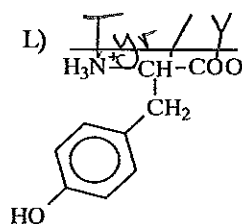
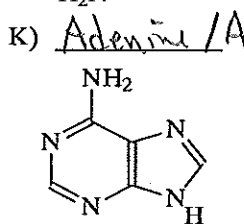
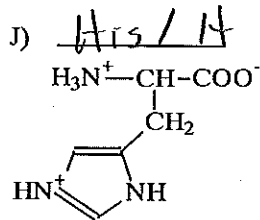
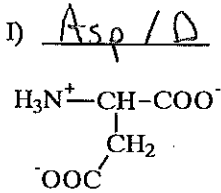
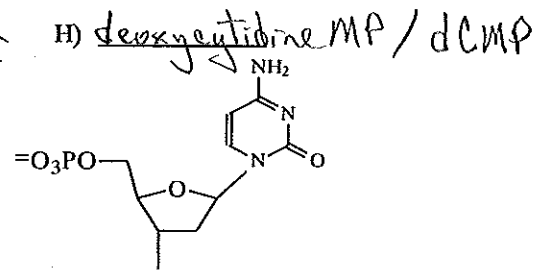
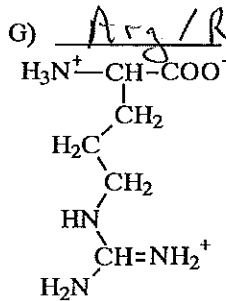
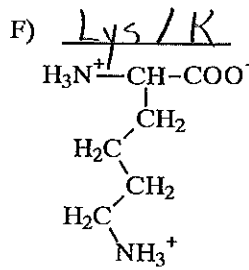
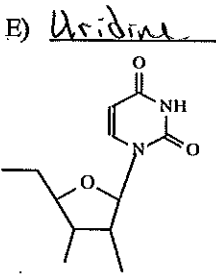
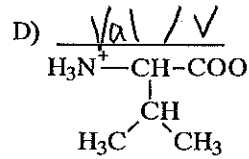
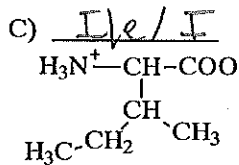
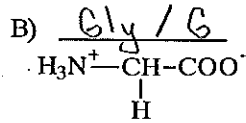
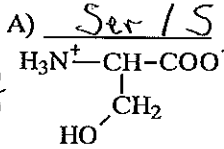


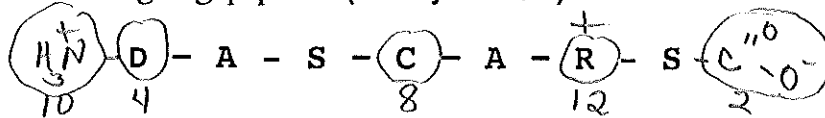
Given: $c = 3.0 \times 10^{10}$ cm/sec; $k = 1.38 \times 10^{-23}$ J/K; $h = 6.63 \times 10^{-34}$ J-sec; $N_0 = 6.02 \times 10^{23}$ /m
 $|E(1\%) \times MW| = |10 \times \epsilon|$; $\text{Eff.} = 1 / (1 + (R/R_0)^6)$; $E = h\nu$

1. Identify each amino acid by its three and one letter codes (e.g. Ala / A, etc.) and each N base, nucleoside or nucleotide by its full name in the blanks above or beside the structures (22 pts).

1/2 pt
each half



2. Consider the following oligopeptide: (show your work)



a) How many titratable protons are there in this oligopeptide under normal conditions? 5 (1/1)
 (2) (Circle the groups involved.)

b) What is the net charge on this oligopeptide at very low pH (pH ~ 1.0)? +2 (2)
 (2)

c) What is the approximate pI for this oligopeptide? 6 (3)
 (3)

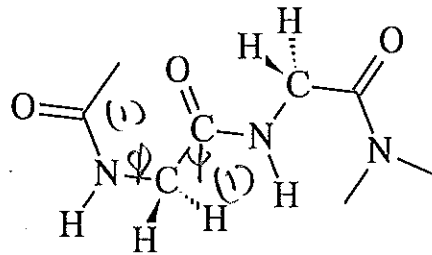
$$+2 \xrightarrow{2} +1 \xrightarrow{4} 0 \xrightarrow{8} -1 \xrightarrow{10} -2 \xrightarrow{12} -3$$

d) Describe how you could take advantage the charge properties resulting from the amino acid content of this oligopeptide to purify it by ion exchange chromatography. Describe a chromatography experiment to enable this oligo to bind and then be eluted from your column by describing the materials and conditions (pH, etc) needed.

(4) You want peptide to bind to column. Since pI ~ 6, run anion exchange col. like DEAE matrix at pH = 8 so peptide binds, then elute with high salt.
 At pH ~ 4, could use cation exch. col. like CME.

3. Label the phi (ϕ) and psi (ψ) angles on the *first glycyl residue* shown below. In the blanks provided, indicate the values of $\phi = \underline{0^\circ}$ and $\psi = \underline{180^\circ}$ for this residue:

(4) (1) (1)



4. A classmate asks for your help understanding the relationships and differences between using Maxam-Gilbert (MG), dideoxy (DD) nucleotides and pyrosequencing of DNA.

a) For each of the following terms, place a "DD" for dideoxy, "MG" for Maxam-Gilbert and/or "PY" depending on which sequencing method the terms apply to. Note: more than one method may apply a term.

(6)

- | | | | |
|--------------------------|---------------|-------------------------|---------------|
| i) use of DNA polymerase | <u>DD, PY</u> | ii) "ladder" sequencing | <u>MG, DD</u> |
| iii) 5'-primer | <u>DD, PY</u> | iv) cleavage reactions | <u>MG</u> |
| v) luciferase | <u>PY</u> | vi) emulsion PCR | <u>PY</u> |

1 pt each if correct
 1/2 partly correct

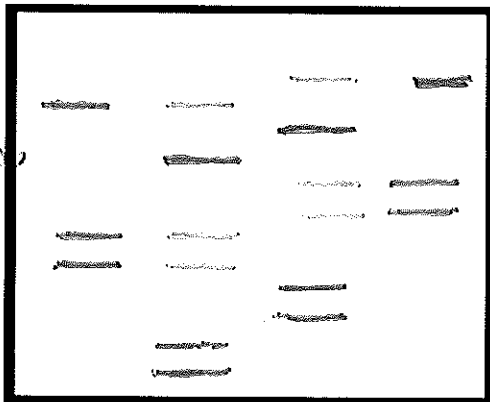
5. DNA structure: When someone discusses the "melting temperature" of DNA, a) what is being measured, and b) what factors would contribute to a higher T_m for a DNA molecule? c) Describe one feature of A-DNA that distinguishes it from B-DNA?

- (6) a) Unstacking of base pairs / unfolding / Abs. inc. as base unstack, T_m is inflection pt. when $\frac{1}{2}$ DNA has unfolded.
- b) More G=C base pairs
- c) A-DNA: C3' endo sugar pucker / $\sim 23\text{\AA}$ dia / 28\AA pitch / major minor groove
 B-DNA: C2' " " " / $\sim 20\text{\AA}$ dia / 34\AA pitch / large major groove

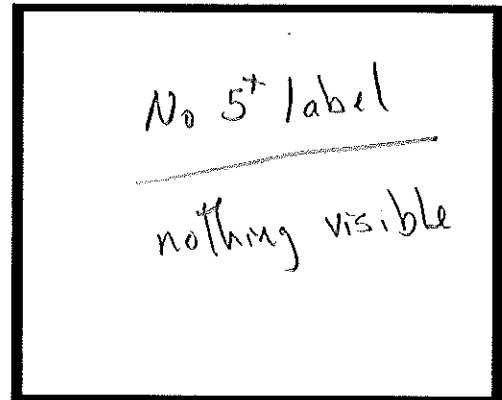
6. Consider the following nucleic acid sample: 5'-AATTGGCCATGC-3' with a 5' radiolabel. On gel "a" below, draw the expected gel pattern that would be obtained if the student carried out Maxam-Gilbert sequencing on this sample. On gel "b" below, draw the expected gel pattern that would occur if the student forgot to add the 5' radiolabel to the oligonucleotide

(4 + 3) a) (1) G rxn (1) G+A rxn (1) T+C rxn (1) C rxn

-2 backwards



b) G rxn G+A rxn T+C rxn C rxn



7. Complete the table below by calculating the specific activities for the following steps used to purify an enzyme, and then answer the questions below: (Show your work)

Step	[Protein] (mg/mL)	Volume (mL)	Total Activity (units)	Specific Activity
1. Crude extract	5.2	285	56,550	38
2. Salt ppt	10.4	38	35,750	90
3. Ion exchange Chrom	7.5	26	31,250	160
4. Affinity Chrom.	2.6	4.2	23,400	2143
5. Gel filtration	1.9	4.8	21,600	2368

(3) Fill in the table with the specific activity values.

- (2) a) What is the overall percentage "yield" for this purification scheme? $\frac{3870}{56550} \times 100 = 6.84\%$
- (2) b) What is the number fold improvement for this purification scheme? $\frac{2368}{38} = 62.3$

