

X-Ray Crystallography

"If a picture is worth a thousand words, then a macromolecular structure is priceless to a physical biochemist." – van Holde

Topics:

1. Image Formation (*optical illusions*)

Resolution / Wavelength (Amplitude, Phase) / Light Microscopy / EM / X-ray / (NMR)

2. Protein Data Bank (PDB)

Data mining and Protein Structure Analysis Tools

3. X-Ray Crystallography

a) 100 years of X-ray Crystallography

b) Crystal Growth – Materials / Methods

c) Crystal Lattices - Lattice Constants / Space Groups / Asymmetric Unit

➔ d) X-ray Sources – Sealed Tube / Rotation Anode / Synchrotron

➔ e) Data Collection – Methods / Detectors / Structure Factors

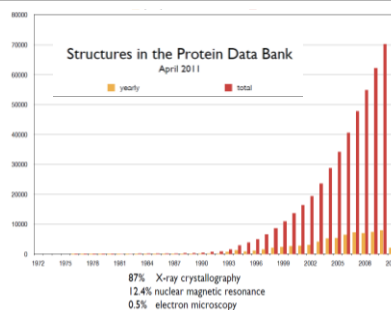
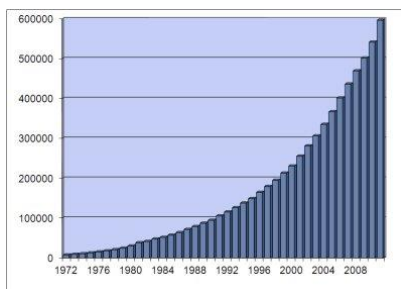
➔ f) Theory of Diffraction – Bragg's Law / Reciprocal Space

➔ g) Structure Solution – Phase Problem: MIR / MR / MAD

➔ h) Refinements and Models / Analysis and presentation of results

A Half Century of Advances in Small Molecule Crystallography

Decade	1950's	1960's	1970's	1980's	1990's	2000's
Size	30 atoms	<100 non-H atoms	<200 non-H atoms	<400 non-H atoms	<600 non-H atoms	<1000 non-H atoms
Time / Structure	~ 1year	3-6 months	1 month	1 week	Few days	Few hours
# Structures	<500	~8300	32,000	95,000	229,000	528,000



What happened? 1913 – 1963 vs. 1963 - 2013

1) Sources

2) Instrumentation -

diffractometers / computers / detectors

3) Software / Computers -

FORTTRAN programming → SHELX, CCP4, Phenix, etc.

4) Molecular Biology - cloning / expression systems; sequencing

5) Automation - robotics

6) Methods -

Phasing: Patterson / Heavy atom; MIR; SIR; MR; MAD; SAD

Model Building: Contour tracing, Richards Box, FRODO, COOT

Refinement and Validation

X-ray Sources:

X-ray tubes: the “sealed” tube

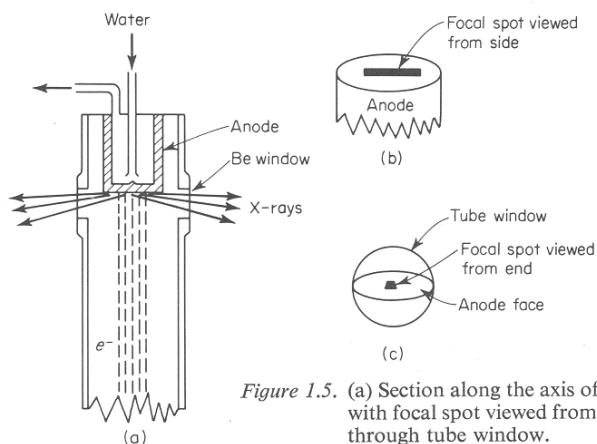
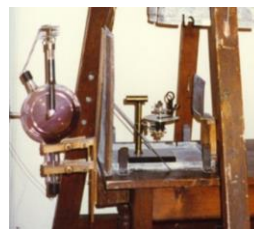


Figure 1.5. (a) Section along the axis of an X-ray tube. (b) Anode with focal spot viewed from side. (c) Focal spot viewed through tube window.



Origin of **Non-characteristic X-rays**

Bremsstrahlung X-rays

In an X-ray tube the electrons emitted from the anode are accelerated towards the metal target cathode by an accelerating voltage of typically 50 kV. The high energy electrons interact with the atoms in the metal target. Sometimes the electron comes very close to a nucleus in the target and is deviated by the electromagnetic interaction. In this process, which is called bremsstrahlung (**braking radiation**), the electron loses much energy and a photon (X-ray) is emitted. The energy of the emitted photon can take any value up to a maximum corresponding to the energy of the incident electron.

The electron (much lighter than the nucleus) comes very close to the nucleus and the electromagnetic interaction causes a deviation of the trajectory where the electron loses energy and an X-ray photon is emitted.



Origin of **characteristic X-rays**

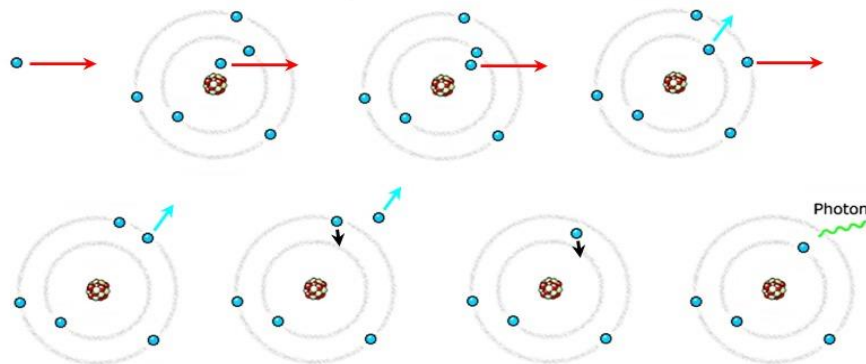
Characteristic X-ray Lines

The high energy electron can also cause an electron close to the nucleus in a metal atom to be knocked out from its place. This vacancy is filled by an electron further out from the nucleus. The well defined difference in binding energy, characteristic of the material, is emitted as a monoenergetic photon. When detected this X-ray photon gives rise to a characteristic X-ray line in the energy spectrum. C. Barkla observed these lines in 1908-09 and was given the 1917 Nobel Prize for this discovery. He also made the first experiments suggesting that the X-rays are electromagnetic waves.

Related Laureate



The Nobel Prize in Physics 1917 - Charles Glover Barkla »



Characteristic X-rays arise from electronic transitions

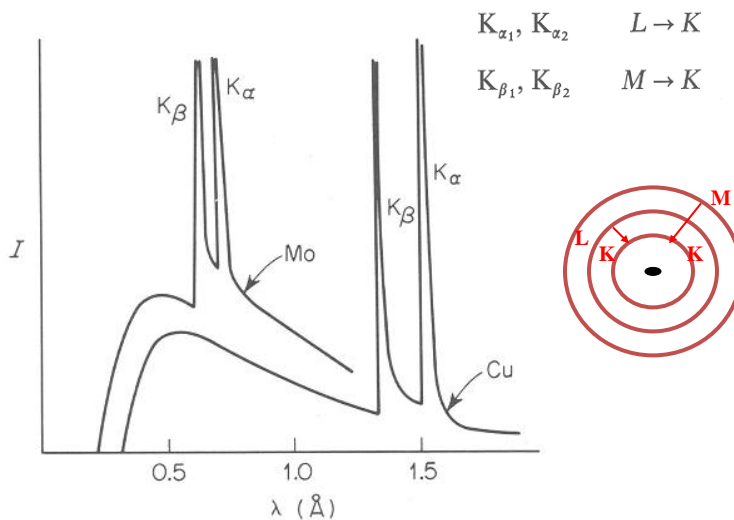


Figure 1.2. X-ray spectra with characteristic peaks: MoK α , 50 Kv; CuK α , 35 Kv.

Characteristic X-rays have defined λ

Table 1.1. Target Materials and Associated Constants

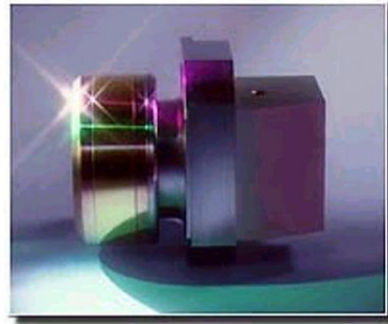
	Cr	Fe	Cu	Mo
Z	24	26	29	42
$\alpha_1, \text{\AA}$	2.2896	1.9360	1.5405	0.70926
$\alpha_2, \text{\AA}$	2.2935	1.9399	1.5443	0.71354
$\bar{\alpha},^* \text{\AA}$	2.2909	1.9373	1.5418	0.71069
$\beta_1, \text{\AA}$	2.0848	1.7565	1.3922	0.63225
β , filt.	V, 0.4 mil†	Mn, 0.4 mil	Ni, 0.6 mil	Nb, 3 mils
α , filt.	Ti	Cr	Co	Y
Resolution, \AA	1.15	0.95	0.75	0.35
Critical potential, kV	5.99	7.11	8.98	20.0
Operating conditions, kV:	30–40	35–45	35–45	50–55
half- or full-wave-	10	10	20	20
rectified, mA				
constant potential, mA	7	7	14	14

* $\bar{\alpha}$ is the intensity-weighted average of α_1 and α_2 and is the figure usually used for the wavelength when the two lines are not resolved.

† 1 mil = 0.001 inch = 0.025 mm.

X-ray Generators

FR591



FR591 Rotating Anode X-ray Generator

The Nonius' FR591 rotating anode X-ray generator now has dramatically improved the performance of the anode, by a complete redesign. We now have a static shaft and a rotating anode, instead of rotating both. The cooling water flow has also been redesigned to give much higher throughput, higher flow and higher turbulence, which results in better heat transfer and hence better cooling capacity.

Now with the new ULTRA anode you can get 6 kW on a 0.3mm focus!

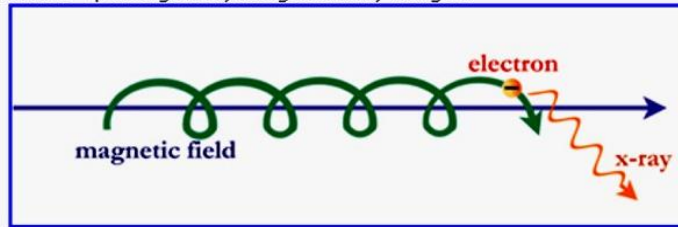


Another Source of "X-rays"

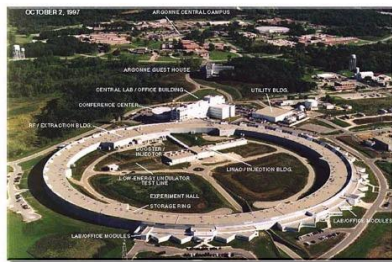
Synchrotron Radiation

X-ray photons can also be created under different conditions. When physicists were operating the first particle accelerators, they discovered that electrons can produce photons without colliding at all. This was possible because the magnetic field in the accelerators was causing the electrons to move in large spirals around magnetic field lines of force. This process is called synchrotron radiation.

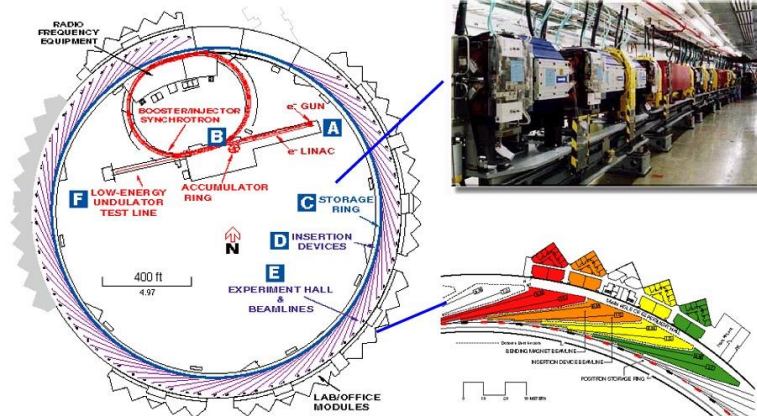
In the cosmos particles such as electrons can be accelerated to high energies—near the speed of light—by electric and magnetic fields. These high-energy particles can produce synchrotron photons with wavelengths ranging from radio up through X-ray and gamma-ray energies.



Synchrotron Radiation: Electrons moving in magnetic field radiate photons.



APS - Advanced Photon Source
Argonne National Laboratory



"X-ray" Sources: X-ray tubes

The **brilliance** of a light source is defined as the number of photons emitted per second, per unit source size, per unit space angle and for a bandwidth of 1/1000 of the photon energy

The Comparison between various sources of X-rays shows large differences in their brilliance.

X-ray tubes:

Wilhelm Conrad Röntgen discovered X-rays in 1895 whilst working with cathode-ray tubes. Using the principle of fast electrons hitting a metallic target, a first substantial gain in brilliance was not obtained until the introduction of rotating anode sources (~1960).

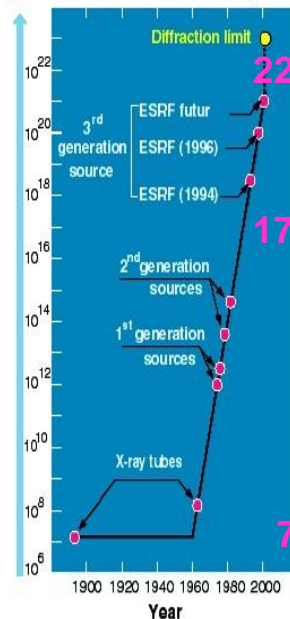
Synchrotron Radiation Facilities:

The progress of high energy physics, with the construction of powerful particle accelerators gave birth to what we now call *First generation* synchrotron sources (~1970). Using the deflection of high energy electrons by a magnetic field for the production of X-rays proved so promising that a number of dedicated *Second generation* sources were built (~1980). Relying on the combination of needle thin electron beams and Insertion Devices, *Third generation* synchrotron sources (~1995) are now emitting synchrotron X-ray beams that are a trillion (10^{12}) times more brilliant than those produced by X-ray tubes.

Free Electron X-ray Lasers:

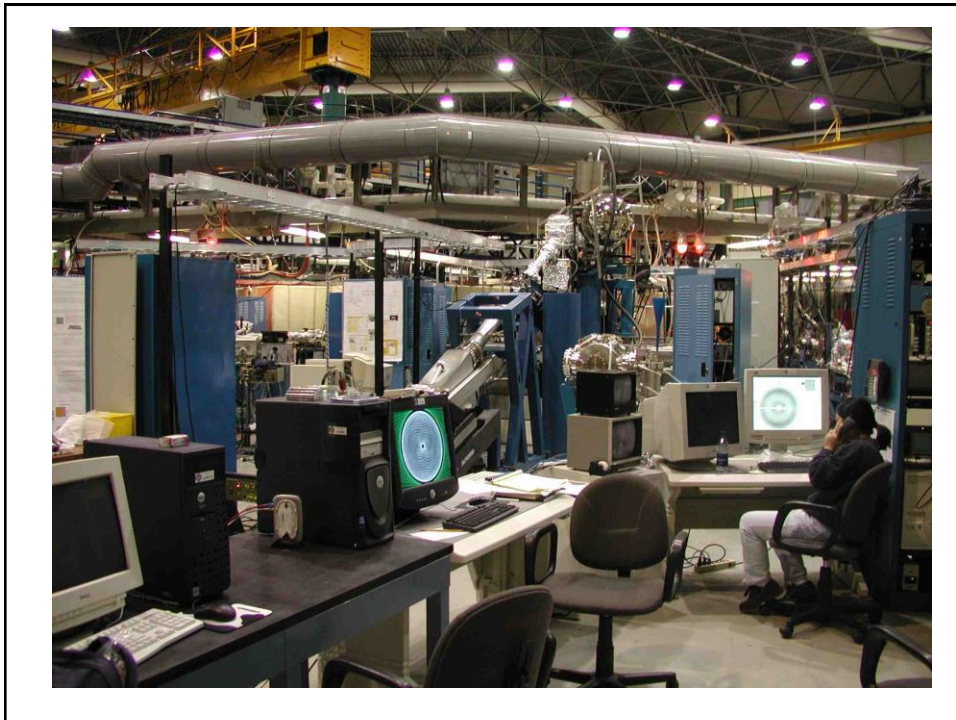
Coupling electron and X-ray beams together, the Free Electron X-ray Lasers currently on the drawing boards could be the next generation of X-ray sources. While they promise to achieve an increase in peak brilliance by another factor of a trillion, the first prototypes may be operational around the year 2010.

Brilliance of the X-ray beams
(photons / s / mm² / mrad² / 0.1% BW)



Advanced Photon Source (APS) synchrotron near Chicago.

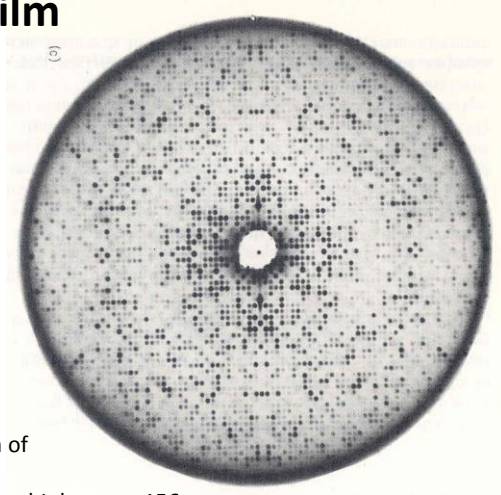




Detectors

Film

Visual intensity estimation

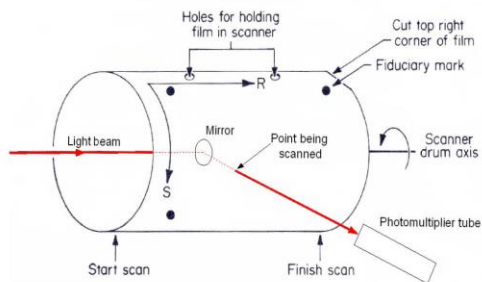


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- 6
- 7
- 8

Precession film of
Phosphorylase
From Blundell and Johnson p.156

Ron Hamlin – Supper Award talk

Concept drawing of a film scanner



Rossmann, Methods in Enzymology 1985, p. 242

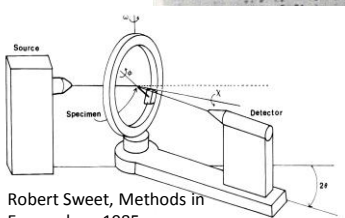
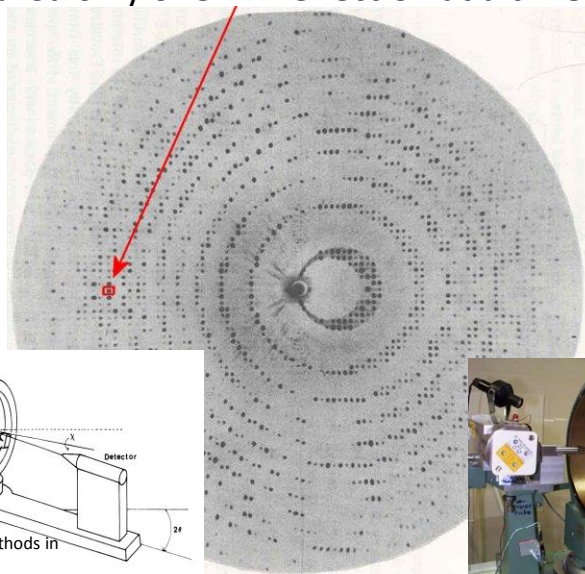
Optronics P-1000 film scanner



Ron Hamlin – Supper Award talk

Photo provided by Dieter Schneider, Brookhaven National Laboratory

Diffractometer: automatic but
measured only **one hkl reflection at a time**

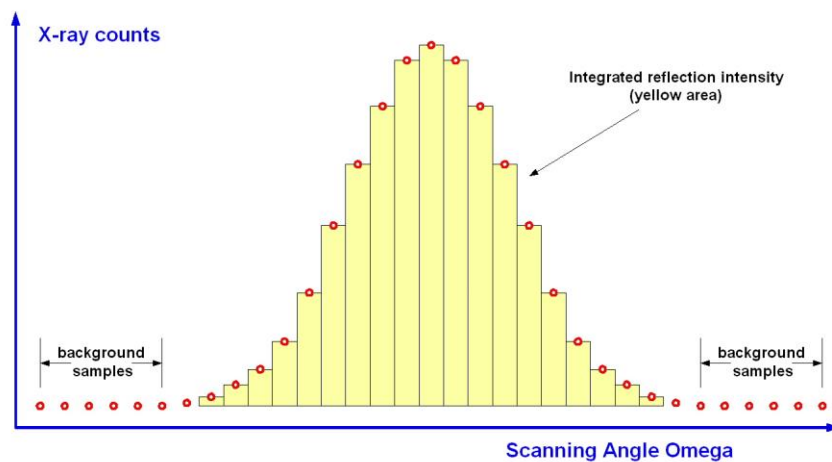


Robert Sweet, Methods in Enzymology 1985



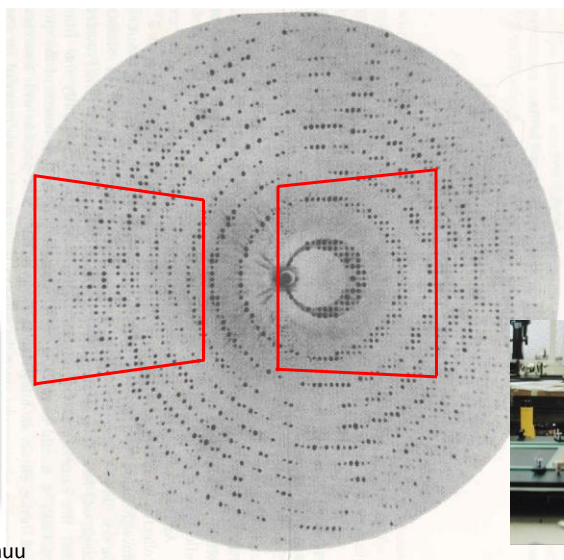
Ron Hamlin – Supper Award talk

Omega scan of a single reflection



Ron Hamlin – Supper Award talk

Area Detectors: Typical coverage of diffraction pattern by a pair of ADSC multiwire detectors

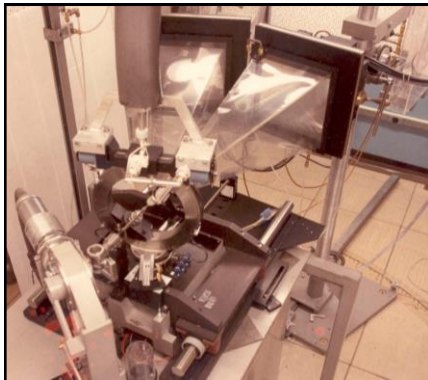


Xuong Nguyen-huu



Ron Hamlin

Ron Hamlin – Supper Award talk



The two-detector Mark II system started operation in Xuong's lab in about 1982

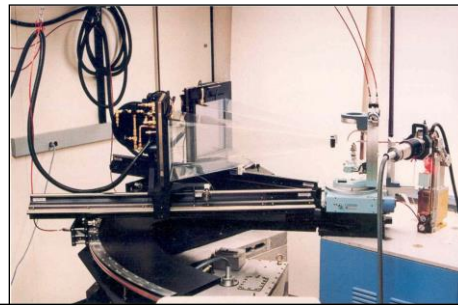
Motor driven two theta table
U. Texas Austin 1988

Ron Hamlin – Supper Award talk

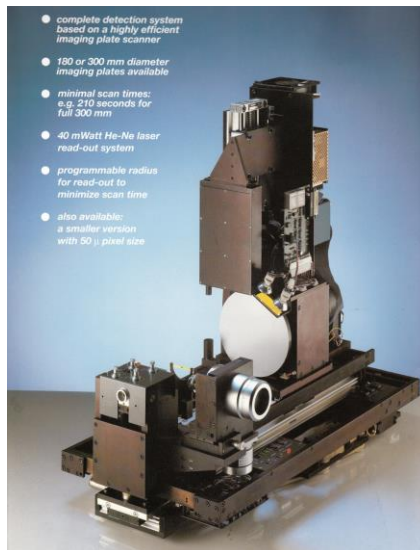
Era of Multiwire Area Detectors



Parts of first ADSC multiwire counter system in Ron's living room in early 1984



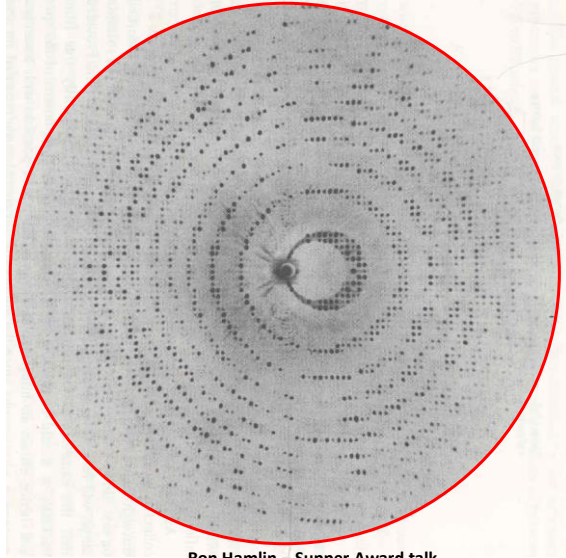
MAR 180 with cover removed



Ron Hamlin – Supper Award talk

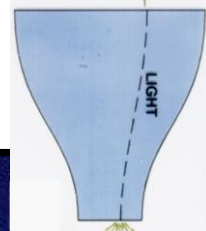
Image Plate Detectors

brute force solid angle coverage



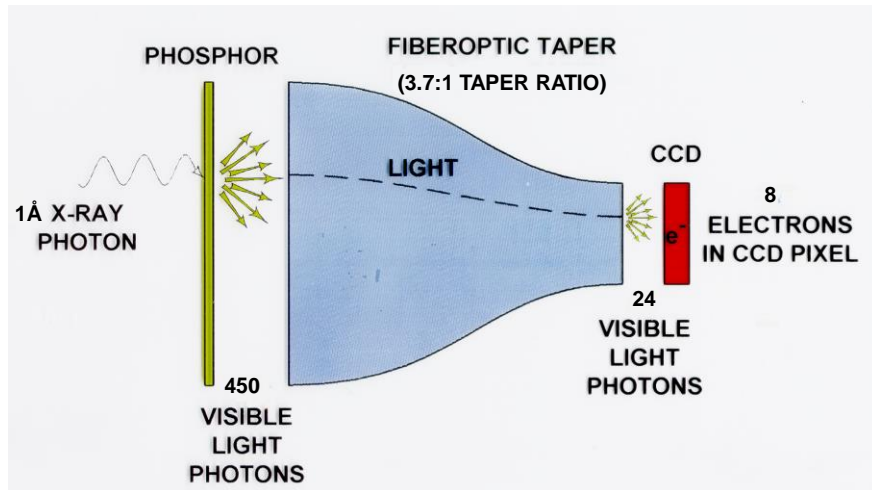
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Fiberoptic Tapers



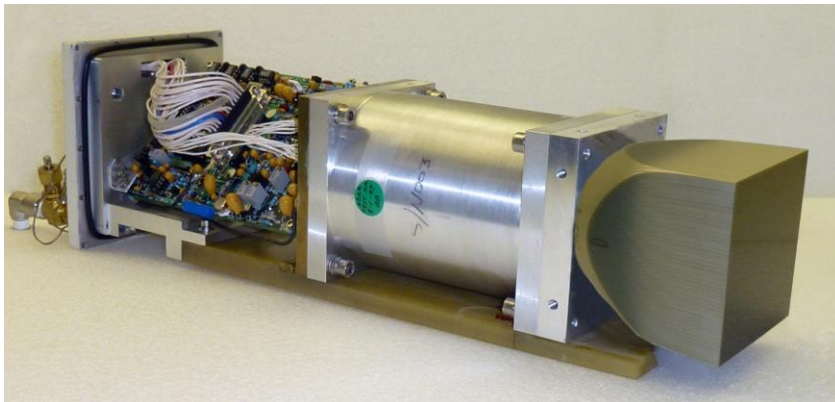
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Basic Principle of Operation



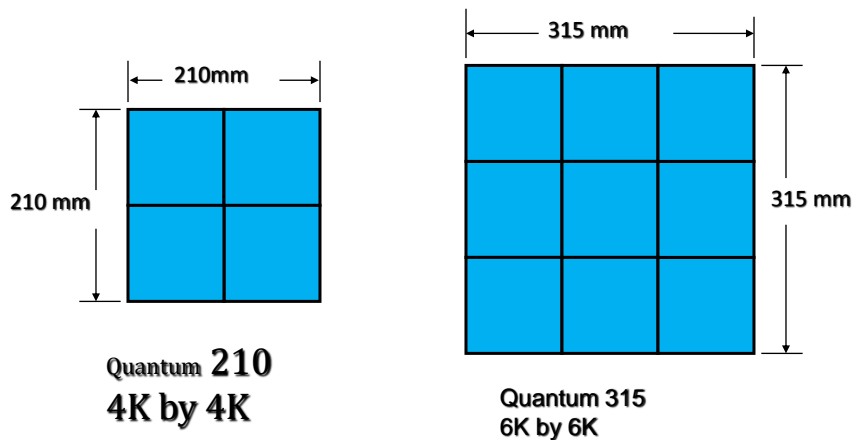
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Quantum 1 cover removed



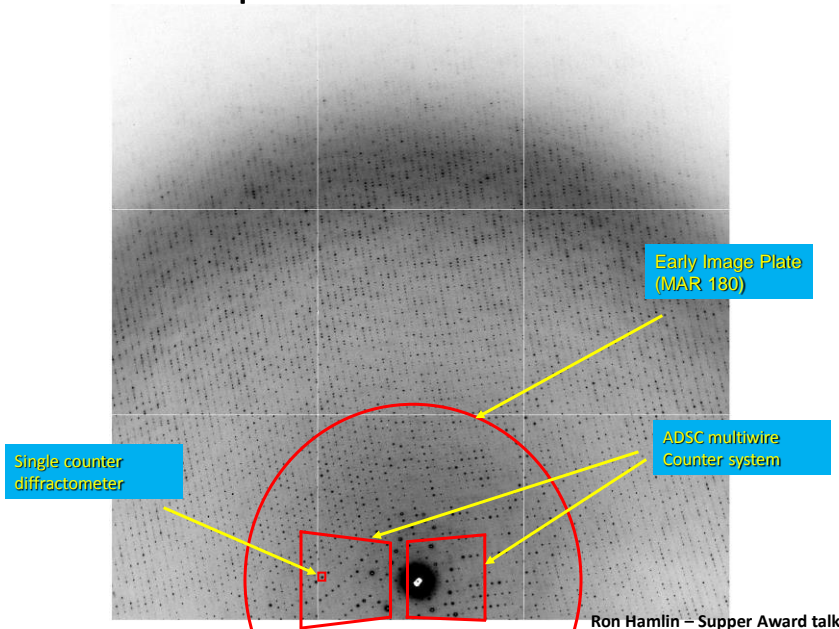
Ron Hamlin – Supper Award talk

The Quantum 315 uses 9 instead of 4 of exactly the same modules as used in the Quantum 210.



Ron Hamlin – Supper Award talk

Diffraction pattern from a Quantum 315



Ron Hamlin – Supper Award talk

X-Ray Crystallography

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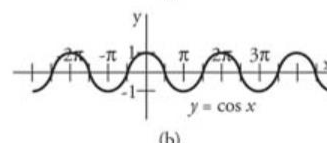
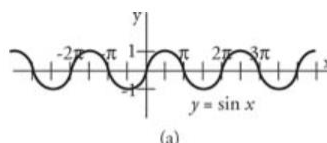
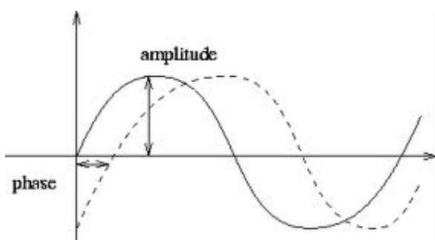
e) Data Collection – Methods / Detectors / Structure Factors

→ f) Theory of Diffraction – Waves, Fourier / Bragg's Law / Reciprocal Space

→ g) Structure Solution – Phase Problem: MIR / MR / MAD

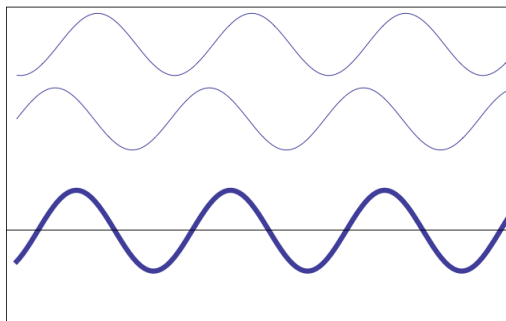
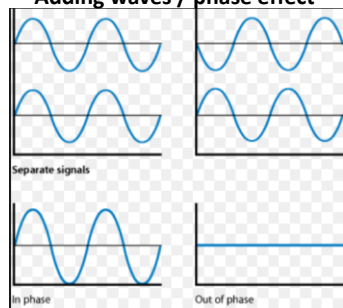
→ h) Refinements and Models / Analysis and presentation of results

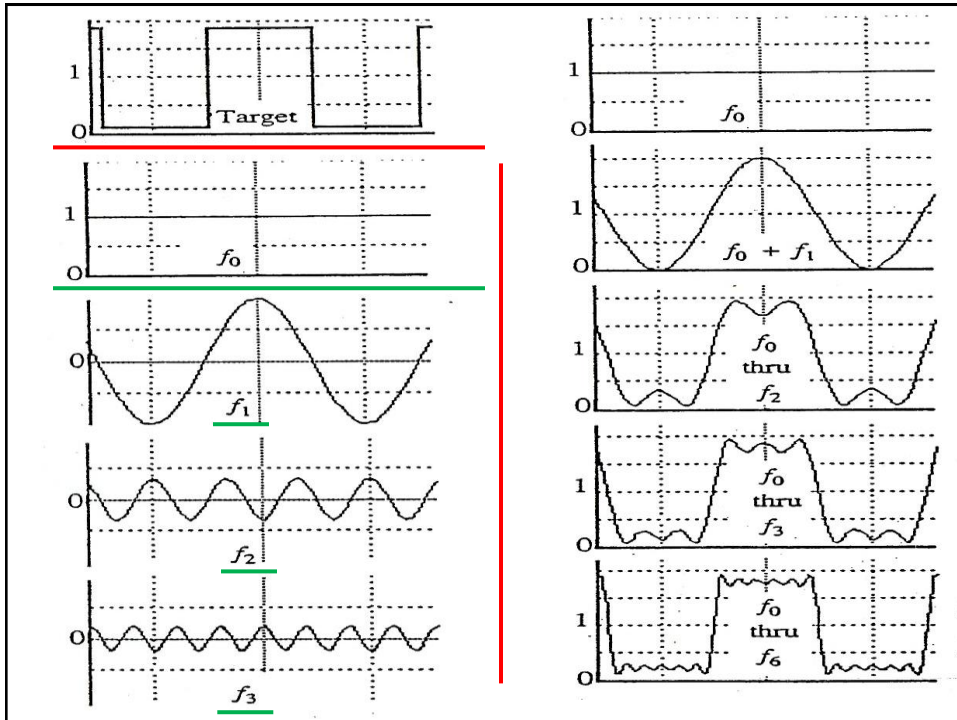
Waves (Amplitudes & Phases)



$$\sin^2 \theta + \cos^2 \theta = 1$$

Adding waves / phase effect





Joseph Fourier / Fourier Series ~1808



Fourier series are named in honor of Joseph Fourier (1768-1830), who made important contributions to the study of trigonometric series, after preliminary investigations by Euler, d'Alembert, and Bernoulli. He applied this technique to find the solution of the heat equation, publishing his initial results in 1807, and publishing his *Théorie analytique de la chaleur* in 1822.

$$f(t) = \frac{a_0}{2} + \sum_{n=1}^{\infty} a_n \cos \frac{n\pi t}{L} + \sum_{n=1}^{\infty} b_n \sin \frac{n\pi t}{L}$$

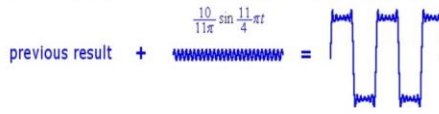
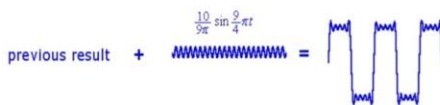
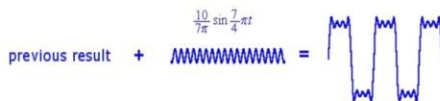
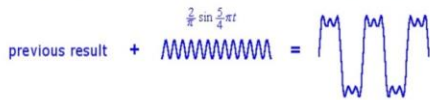
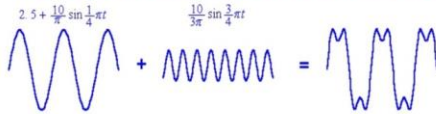
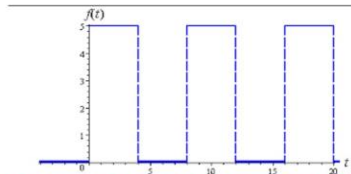
$$a_0 = \frac{1}{L} \int_{-L}^L f(t) dt$$

$$a_n = \frac{1}{L} \int_{-L}^L f(t) \cos \frac{n\pi t}{L} dt \quad b_n = \frac{1}{L} \int_{-L}^L f(t) \sin \frac{n\pi t}{L} dt$$

where $n = 1, 2, 3 \dots$

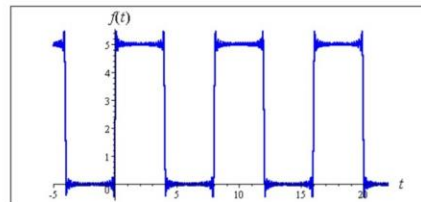
Fourier Series

Example - Square Wave



If we graph many terms, we see that our series is producing the required function. We graph the first 20 terms:

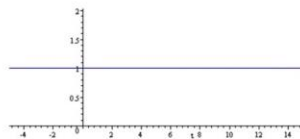
$$2.5 + \frac{10}{\pi} \sum_{n=1}^{20} \frac{1}{(2n-1)} \sin \frac{(2n-1)\pi t}{4}$$



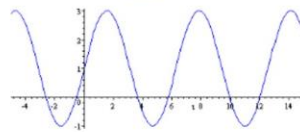
Example - Saw Tooth Function

$$f(t) = 1 + 2 \sin t - \sin 2t + \frac{2}{3} \sin 3t$$

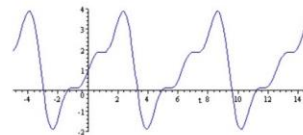
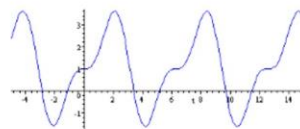
$f(t) = 1$ (first term of the series):



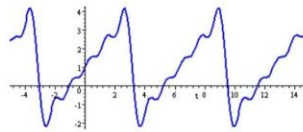
$f(t) = 1 + 2 \sin t$ (first 2 terms of the series):



$f(t) = 1 + 2 \sin t - \sin 2t$ (first 3 terms of the series):

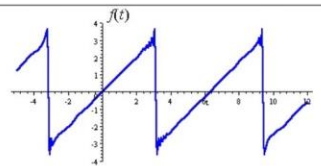


$$f(x) = 1 + 2 \sin t - \sin 2t + \frac{2}{3} \sin 3t - \frac{1}{2} \sin 4t + \frac{2}{5} \sin 5t + \dots$$



The graph of the first 40 terms is:

$$\sum_{n=1}^{40} \left(\frac{2}{n} (-1)^{n+1} \sin nt \right)$$



Fourier Series Applet <http://www.falstad.com/fourier/>

Sines / Cosines / and Exponentials

$$\begin{aligned}\exp(x) &\equiv e^x \equiv \sum_{n=0}^{\infty} \frac{x^n}{n!} \\ &= 1 + x + \frac{x^2}{2} + \frac{x^3}{6} + \frac{x^4}{24} + \frac{x^5}{120} + \dots\end{aligned}$$

If we let x be *imaginary*, $x = i\theta$ (where θ is real), then this can be written

$$e^{i\theta} = 1 + i\theta - \frac{\theta^2}{2} - i\frac{\theta^3}{6} + \frac{\theta^4}{24} + i\frac{\theta^5}{120} - \dots$$

recall

$$\cos \theta = 1 - \frac{\theta^2}{2} + \frac{\theta^4}{24} - \dots$$

\mathcal{E}

$$\sin \theta = \theta - \frac{\theta^3}{6} + \frac{\theta^5}{120} - \dots$$

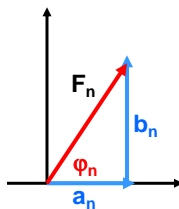
thus

$$e^{i\theta} = \cos \theta + i \sin \theta$$

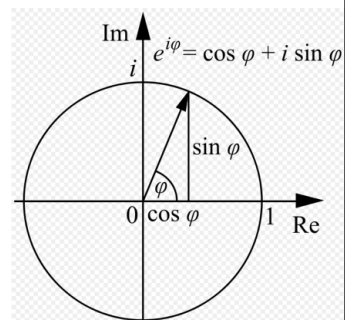
Euler's Equation

Euler's formula (Leonhard Euler, 1707-1783) gives the relationship between the complex exponential function and common trig terms. For any real number " φ "

→ Replace list of a_n / b_n with $|F_n|$ and φ_n



n	a	b	F	φ
1	7	0	7	0
2	0	8	8	90
3	5	5	7.1	??
4	8	6	??	36.9



Fourier Series

 /

Fourier Transforms

$$f(t) = \frac{a_0}{2} + \sum_{n=1}^{\infty} a_n \cos \frac{n\pi t}{L} + \sum_{n=1}^{\infty} b_n \sin \frac{n\pi t}{L}$$

or

$$a_0 = \frac{1}{L} \int_{-L}^L f(t) dt$$

$$a_n = \frac{1}{L} \int_{-L}^L f(t) \cos \frac{n\pi t}{L} dt \quad b_n = \frac{1}{L} \int_{-L}^L f(t) \sin \frac{n\pi t}{L} dt$$

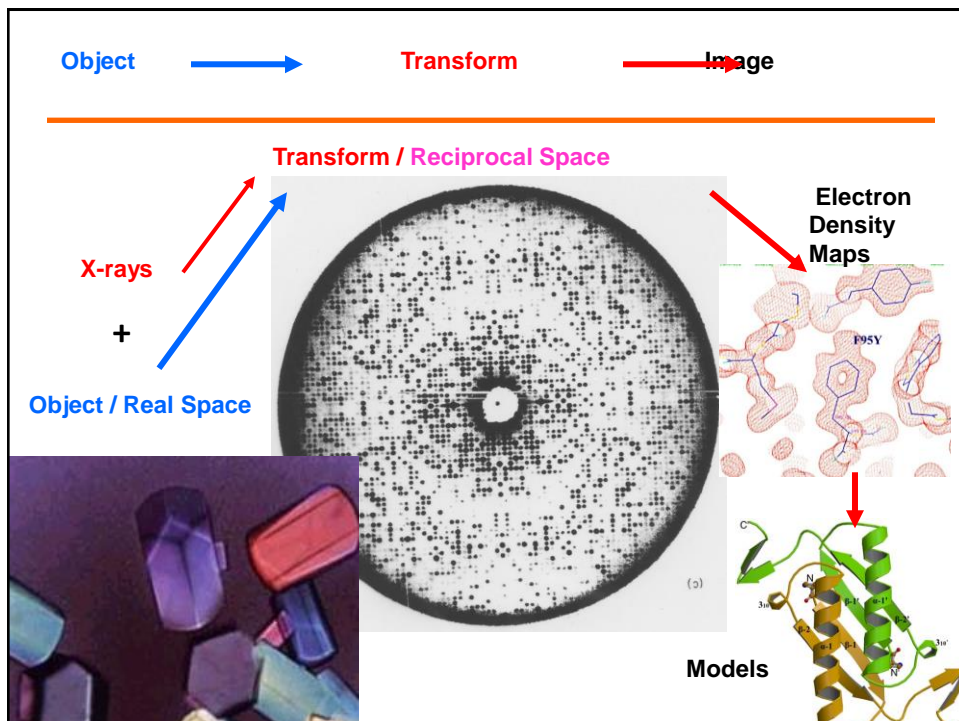
Now consider **electron density** (as a **function** or a set of **coefficients**)

$$\rho(\mathbf{x}) = \frac{1}{V} \sum_{\mathbf{h}} \mathbf{F}(\mathbf{h}) \exp(-2\pi i \mathbf{h} \cdot \mathbf{x}) \quad \text{or} \quad \mathbf{F}(\mathbf{h}) = \int \rho(\mathbf{x}) \exp(2\pi i \mathbf{h} \cdot \mathbf{x}) d\mathbf{v}$$

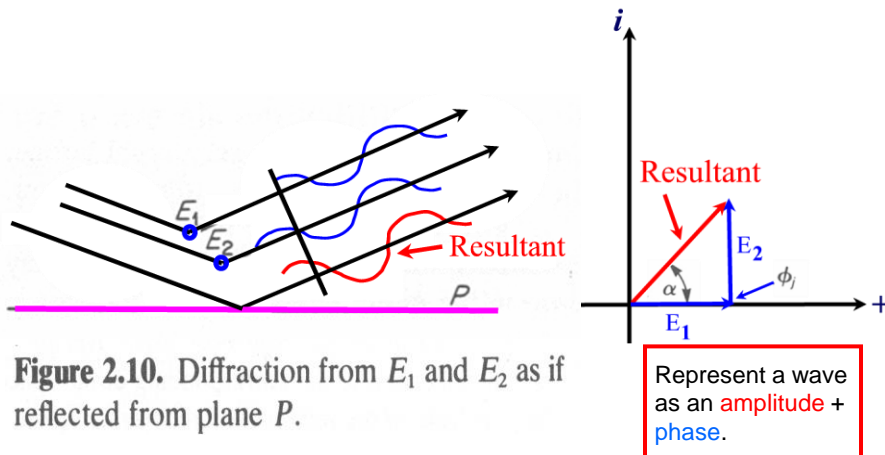
Real Space Reciprocal Space cell

AND – \mathbf{F}_{hkl} can **also** be **calculated** as the resultant scattering or the **sum of the individual scattering atoms!!**

$$\mathbf{F}(hkl) = F(hkl) e^{i\alpha(hkl)} = \sum_{j=1}^{N'} \mathbf{f}_j(hkl) = \sum_{j=1}^{N'} f_j(hkl) e^{i\phi_j(hkl)}$$



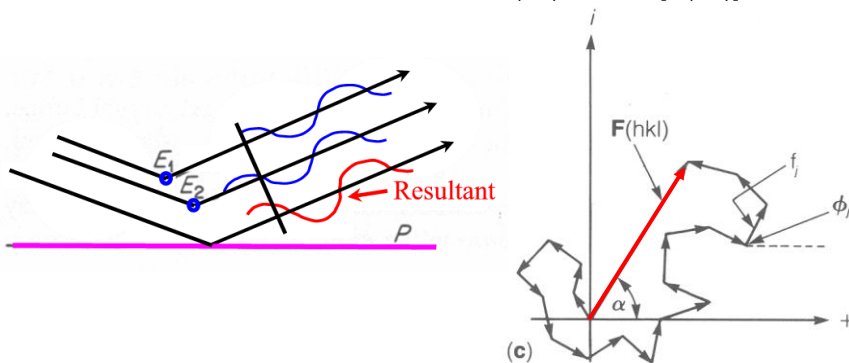
Diffraction: Scattering from (two) “atoms”



Scattering from “many atoms”

$$\mathbf{F}(hkl) = F(hkl)e^{i\alpha(hkl)} = \sum_{j=1}^{N'} \mathbf{f}_j(hkl) = \sum_{j=1}^{N'} f_j(hkl)e^{i\phi_j(hkl)}$$

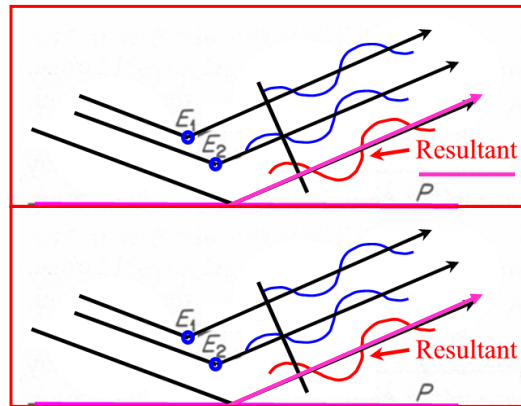
$\mathbf{f}_j(hkl)$ ← Calculated
 $F(hkl)$ ← Experimental



The structure factor for a reflection may be thought of as the vector sum of the x-ray scattering contributions from many atoms.

Each of the j contributions may be represented as a vector in the complex plane, with amplitude f_j and phase ϕ_j .

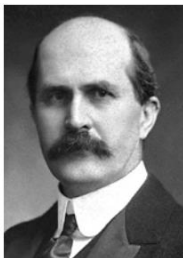
Scattering from “atoms in two unit cells”



The Nobel Prize in Physics 1915
William Bragg, Lawrence Bragg



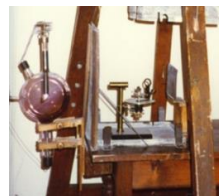
The Nobel Prize in Physics 1915



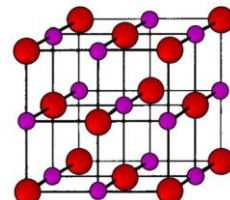
Sir William Henry Bragg



William Lawrence Bragg

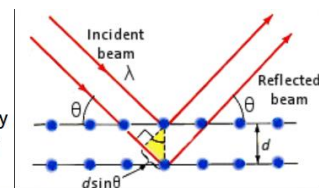


X-ray apparatus



Sodium Chloride (NaCl)

The Nobel Prize in Physics 1915 was awarded jointly to Sir William Henry Bragg and William Lawrence Bragg "for their services in the analysis of crystal structure by means of X-rays"



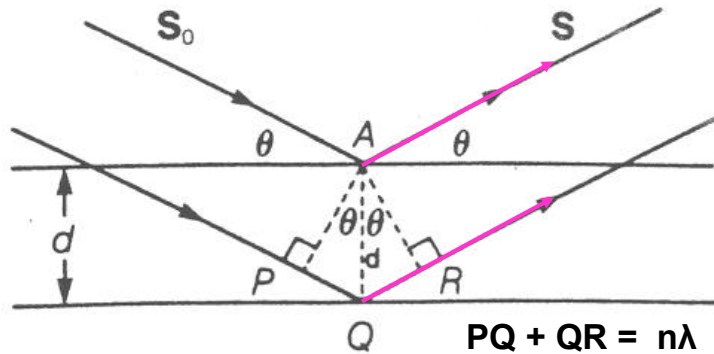
Bragg's Law ($n\lambda = 2d \sin \theta$)

Crystals: Scattering from “planes”

Resultant scattering of resultant scattering!

Bragg Equation

$$n\lambda = 2d \sin(\theta)$$

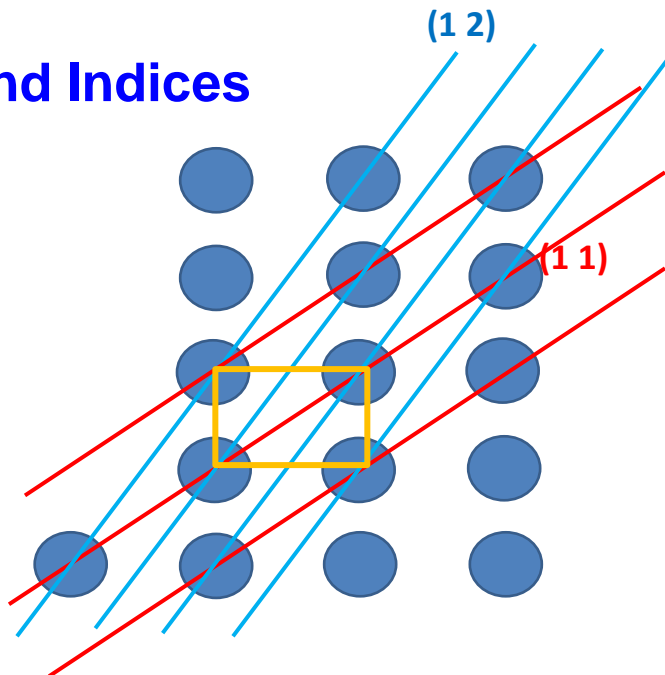


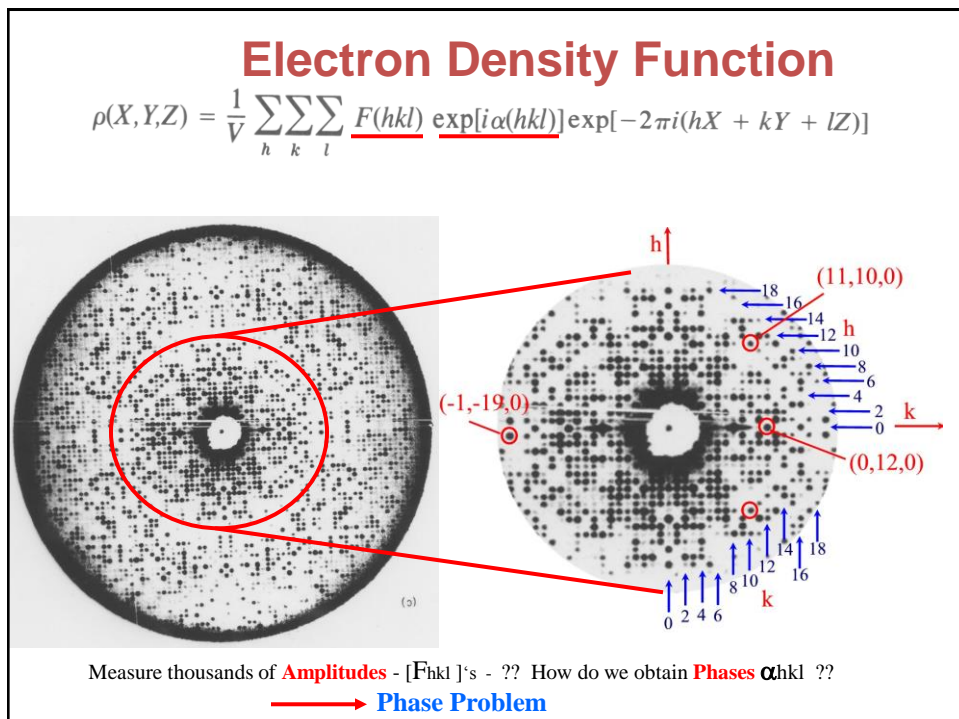
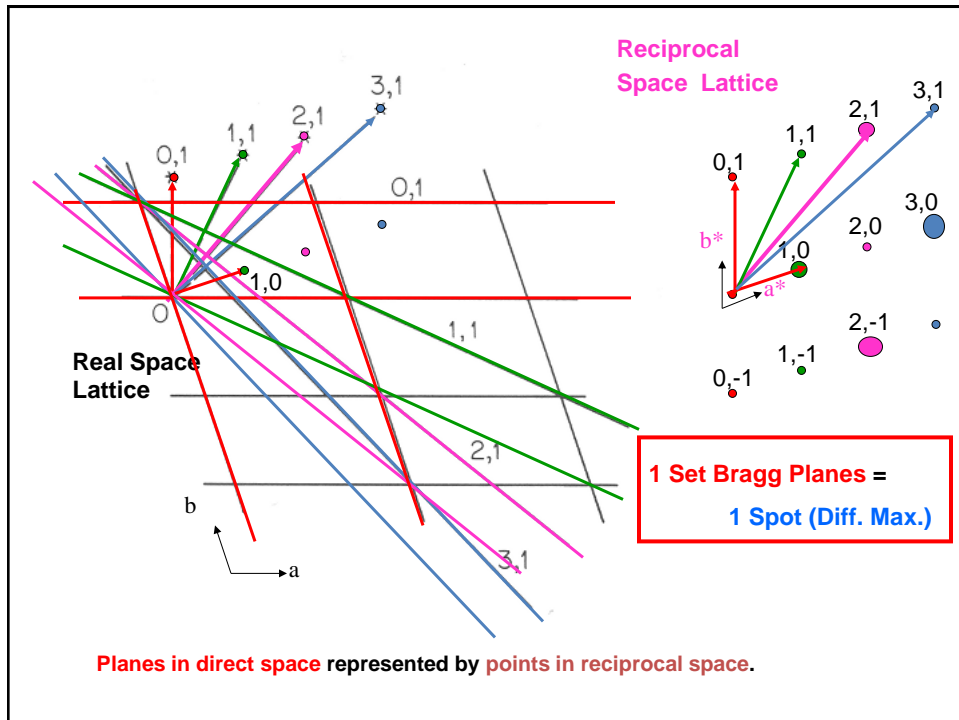
→ Scattering will only be “observed” at discrete Bragg angles(θ)

The spacings of the Bragg reflections

→ Lattice Constants

Planes and Indices

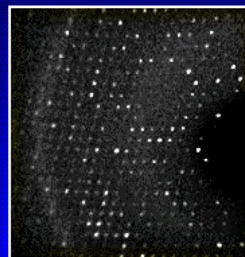
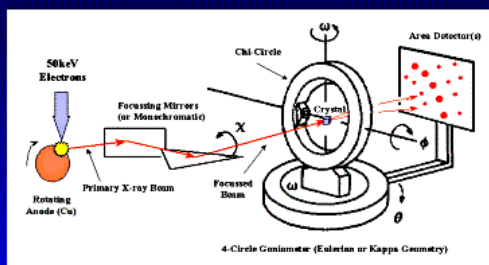




Advanced Methods in Modern Biomolecular Crystallography

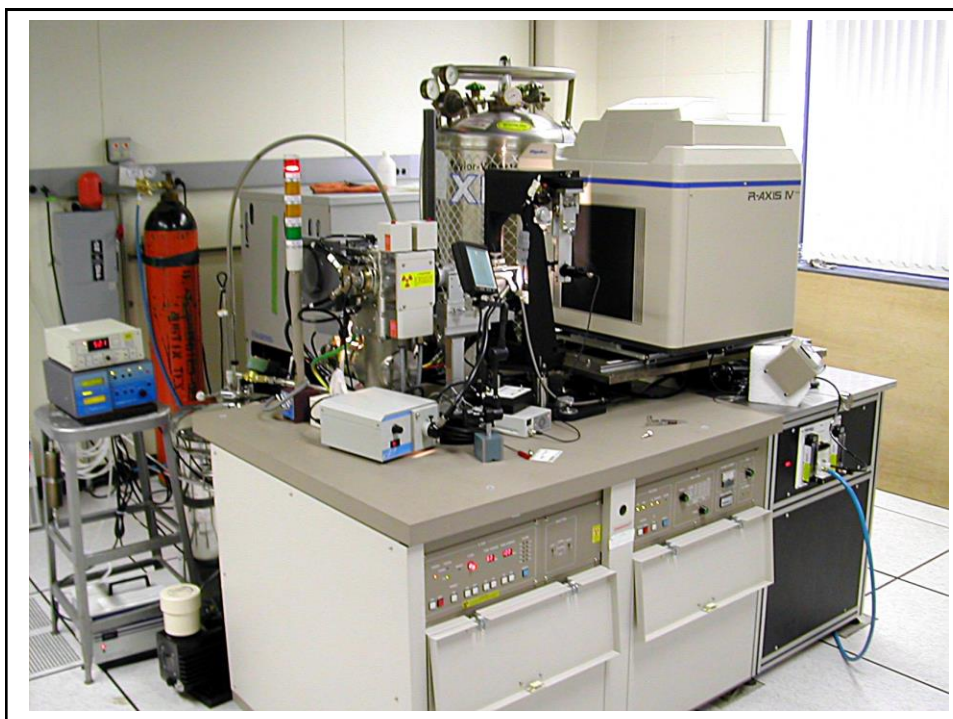


The information we get from a single diffraction experiment.....



The reflections are indexed (consistent assignment of reciprocal cell indices h, k, l) and all we get for the money is a long list of intensities from several ten thousand reflections

2	10	1	326.58
3	10	1	1644.72
4	10	1	3228.45
5	10	1	1279.83
6	10	1	320.48
7	10	1	775.63
8	10	1	1344.55
9	10	1	431.73
10	10	1	1760.14
11	10	1	709.18
12	10	1	20.37
13	10	1	408.72
14	10	1	51.36
15	10	1	114.72
16	10	1	776.26
17	10	1	87.57
18	10	1	30.93
0	11	1	99.30
1	11	1	2258.68
2	11	1	770.18





Cryo-cooling efficiently improves data quality



- Crystals are rapidly cooled (**NOT FROZEN**) to near liquid nitrogen temperature
- Reduced thermal vibrations
- **Increased resolution**
- Reduced disorder
- **Eliminated radiation damage**
- No merging and scaling errors



CrystalClear 3.3.6 [Sample: pvm Instrument: R-AXIS IV+]

File Edit View Sample Instrument Processing Tools Window Help

Tools: Collect and Process

Sample: pvm

Initialize Instrument
Setup
Crystal Evaluation
Mount Crystal
Initial Images
Assign Unit Cell
Find Spots
Index Spots
Pattern Calc
Predict Spots
Strategy
Collect Images
Integrate Reflections
Analyze Data

Collection Image Display - C:\pvm\bin\field\Gel2\pvm\Images\pvm_screen001.png

Contrast: 21
LO: 14 HE: 107
Close

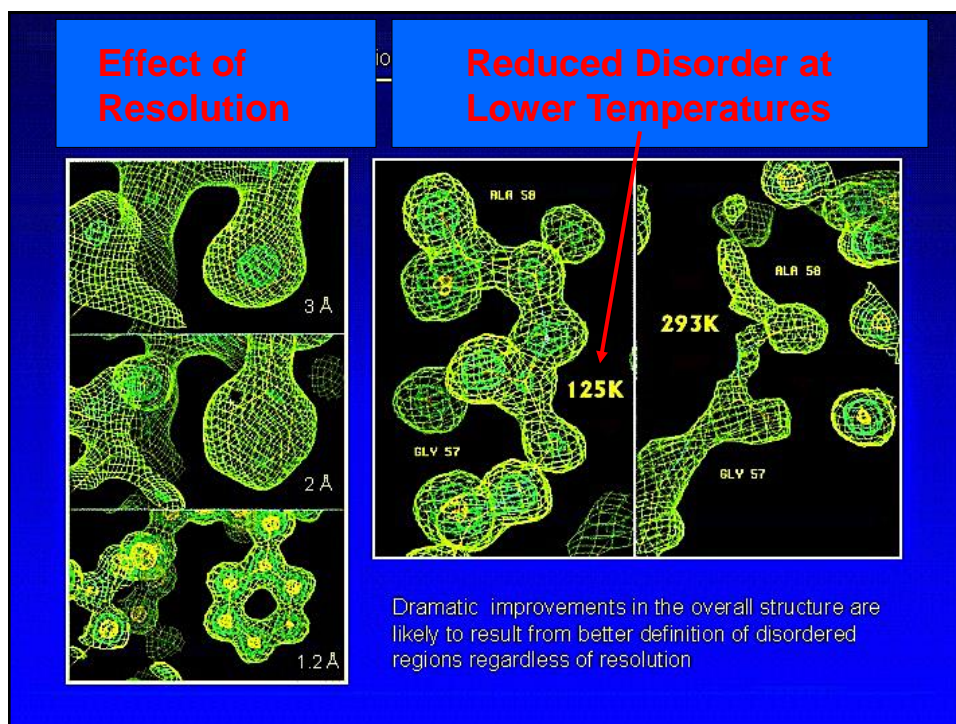
Image name: pvm_screen001
Image number: 1
Reflection list:
Start angle (°): 50.00
Image width (°): 5.50
Exposure time (min): 15.00
Crystal to Det dist (mm): 150.00
Detector 2θ (°): 7.00
Pixel position: 223.6, 312.1
Pixel value: 25
Peak intensity: 13.7
σ: 1.85 Acc: 0.4
Sign: 0.4
ice spots: 0
More spots: 100% (P4) Plot

Instrument State Display - R-AXIS IV+
X-ray source: Status: Not connected to 2-GS
Shutter: Status: Open (21.3 sec)
Voltage (kV): 20.0 Current (mA): 100.0 SR rate: 0.0
Element: Focus: 0.30 Calibration type: 0.3 x 1.5 Doubt
Source type: Göttingen

Expected finish time: pvm_screen001.asc Wed Apr 07 04:50 PM
Scan: Wed Apr 07 04:50 PM
Schedule:

Messages:
There are no images scheduled for collection. Response = OK (04/07/04 16:34:16)
Continue image collection using the collection schedule? This may be undesirable in that you may overwrite previously collected images as well as overwrite results obtained by processing these previously collected images.
Estimated scan collection finish time: Wed Apr 07 04:50 PM
Response = Yes (04/07/04 16:34:33)

CrystalClear 3.3.6



X-Ray Crystallography

"If a picture is worth a thousand words, then a macromolecular structure is priceless to a physical biochemist." – van Holde

Topics:

1. Image Formation (*optical illusions*)

Resolution / Wavelength (Amplitude, Phase) / Light Microscopy / EM / X-ray / (NMR)

2. Protein Data Bank (PDB)

Data mining and Protein Structure Analysis Tools

3. X-Ray Crystallography

- a) **100 years of X-ray Crystallography**
- b) Crystal Growth – Materials / Methods
- c) Crystal Lattices – Lattice Constants / Space Groups / Asymmetric Unit
- d) X-ray Sources – Sealed Tube / Rotation Anode / Synchrotron
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- ➡ g) Structure Solution – Phase Problem: MIR / MR / MAD
- ➡ h) Refinements and Models / Analysis and presentation of results

Solving the Phase Problem

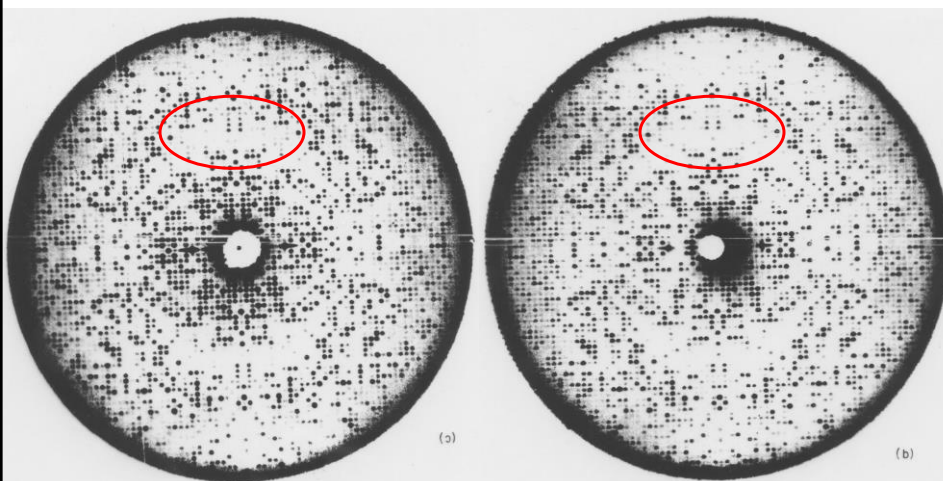
Early Days:

Centric structures (all phases 0° or 180°) Heavy atom / Patterson method

Macromolecular Crystallography

1. **MIR: Multiple Isomorphous Replacement (Heavy Atom)**
 2. **MR: Molecular Replacement**
 3. **MAD: multiwavelength anomalous dispersion**
- *****
- **Molecular Modeling** (predicting starting structure from sequence alone)

Use of Heavy Metal Ions for Phasing by MIR Methods

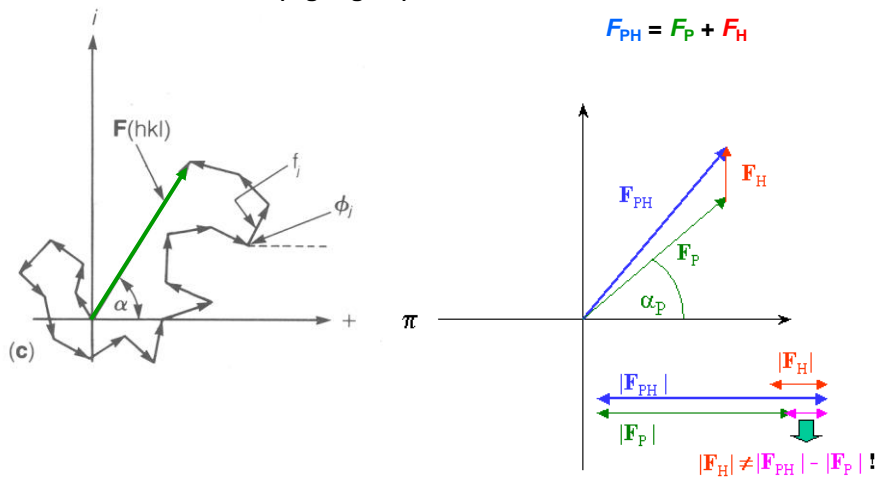


Native Phosphorylase

Phosphorylase + Ethyl Hg
thiosalicylate

Effect of adding 1 “heavy” atom with lots of electrons!

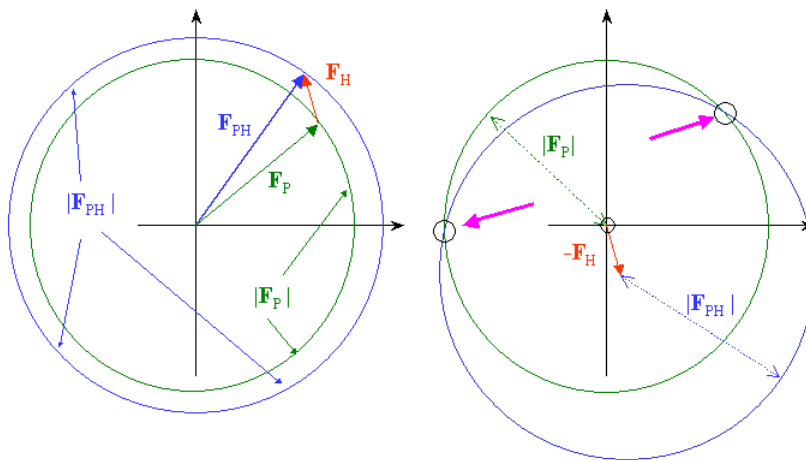
(e.g. Hg, Pt)



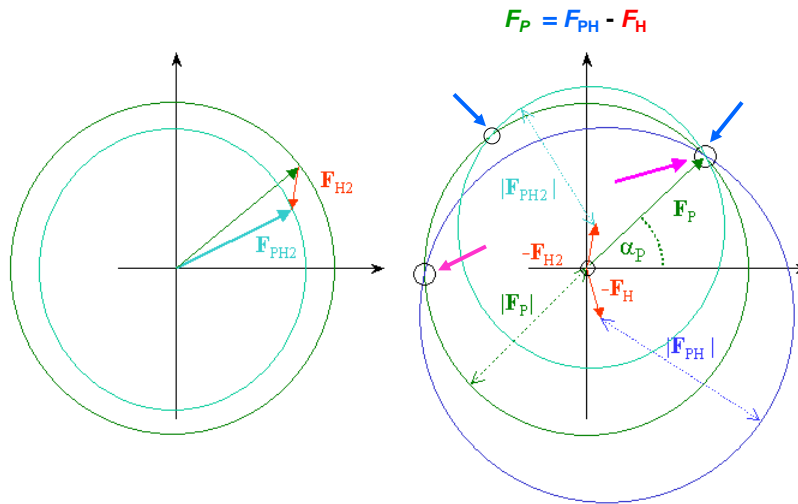
Multiple Isomorphous Replacement (MIR) method

$$F_{PH} = F_P + F_H$$

$$F_P = F_{PH} - F_H$$



Multiple Isomorphous Replacement (MIR) method



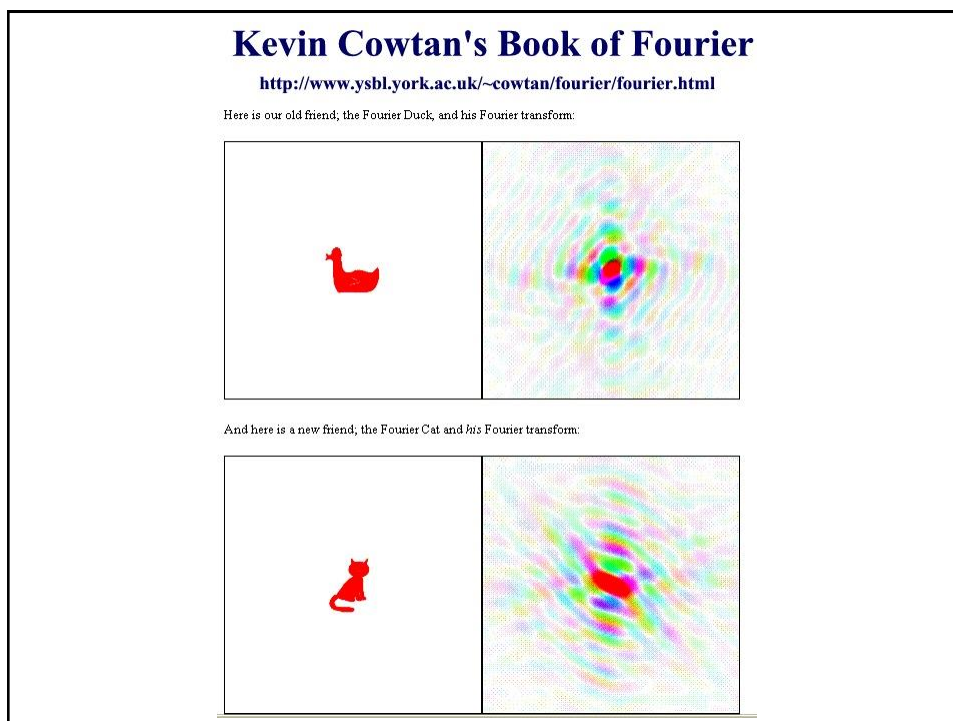
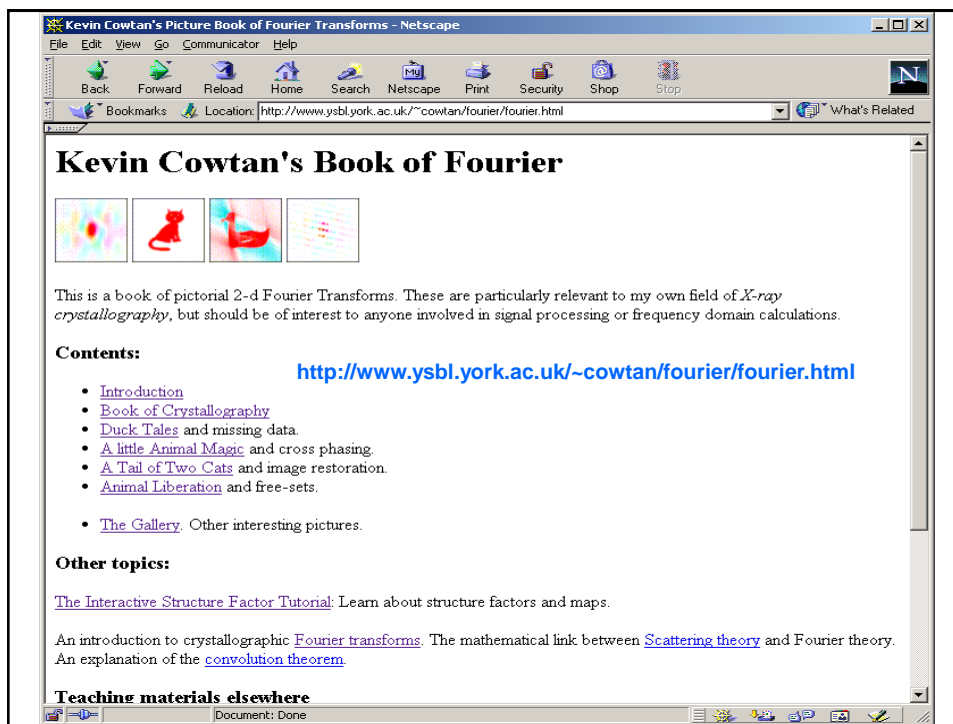
Solving the phase problem by **“Molecular Replacement”**.

If an approximate model of the protein structure is known in advance, approximate phases can be guessed, and the unknown parts of the structure can be calculated in an iterative procedure.

No heavy atom derivative required.

BUT – need starting model and orientation (rotation and translation)

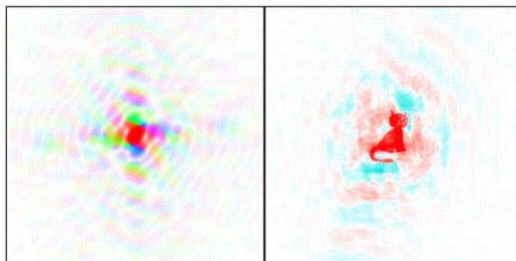
For example, molecular replacement can be used to determine the structure of an **complex with inhibitor** bound to an enzyme active site, if the structure of the enzyme itself is already known. Also, MR is often used to solve the structures of **closely related proteins** in a superfamily.



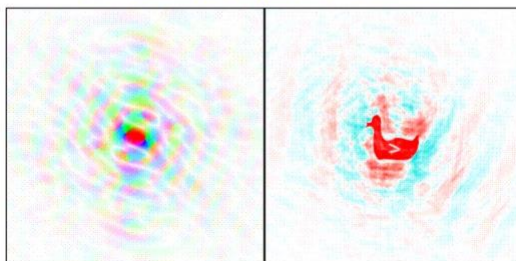
Kevin Cowtan's Book of Fourier

<http://www.yesbl.york.ac.uk/~cowtan/fourier/fourier.html>

Duck Transform Amplitudes + Cat Phases



Cat Transform Amplitudes + Duck Phases



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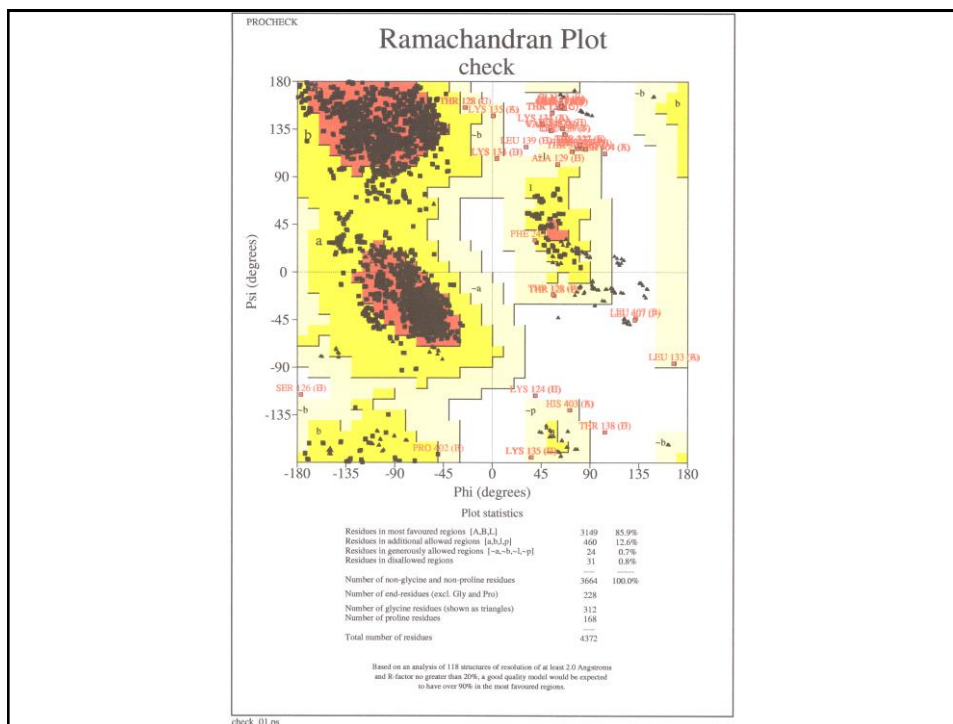
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Crystal Structure of *M. tuberculosis* Alanine Racemase

Table 1: Data Collection and Processing Statistics for the MAD and Native Data Sets of Alr_{Mtb}

	MAD 1	MAD 2	MAD 3	MAD 4	native
λ (Å)	0.9788	0.9790	0.9562	0.9809	0.9160
resolution (Å)		2.20		1.80	
mosaicity		0.50		0.65	
no. of reflections observed $> 1\sigma$	432376	446744	431524	336135	779600
no. of unique reflections $> 1\sigma$	35817	37506	36020	36242	67592
R_{merge}^a (%)	6.9	6.4	5.1	3.7	6.0 (67.2)
completeness (%)	91.8	95.8	92.1	92.1	99.3 (95.6)
$\langle I/\sigma \rangle$	30.3	34.3	41.6	50.9	34.5 (2.6)

$$^a R_{\text{merge}} = \sum |I_{\text{obs}} - I_{\text{avg}}| / \sum I_{\text{avg}}$$

Biochemistry 2005, 44, 1471–1481

1471

The 1.9 Å Crystal Structure of Alanine Racemase from *Mycobacterium tuberculosis* Contains a Conserved Entryway into the Active Site^{1,2}

Pierre LeMuguer,³ Hookang Im,³ Jerry Ebalunode,³ Ulrich Strych,³ Michael J. Benedik,⁴ James M. Briggs,³
Harold Kohn,² and Kurt L. Krause^{1,3,4*}

¹Department of Biology and Biochemistry, University of Houston, Houston, Texas 77204-5001, ²Department of Biology,
Texas A&M University, College Station, Texas 77843-3258, ³Division of Medicinal Chemistry and Natural Products,
School of Pharmacy, University of North Carolina, Chapel Hill, North Carolina 27599-7590, and
⁴Section of Infectious Diseases, Department of Medicine, Baylor College of Medicine, Houston, Texas 77030

Received June 27, 2004; Revised Manuscript Received October 22, 2004

Table 2: Final Refinement Statistics for Alr_{Mtb} at 1.9 Å Resolution

R factor ^a (%)	20.4
R_{free} (%) (for 1747 reflections)	25.4
average B factor (Å ²) ^b	
main chain	25.5
side chain	31.5
PLP	21.9
waters	32.4
rms deviations	
bond lengths (Å)	0.006
bond angles (deg)	1.9
no. of reflections $> 2\sigma$	55001
no. of residues	722
no. of protein atoms	5360
no. of PLP atoms	30
no. of water molecules	350

^a R -factor = $\sum |F_{\text{obs}} - F_{\text{calc}}| / \sum |F_{\text{obs}}|$. ^b All isotropic model.

Energy Refinement

(Simulated Annealing)

$$E_{TOTAL} = E_{EMPIRICAL} + E_{EFFECTIVE}$$

$$E_{EFFECTIVE} = E_{XREF} + E_{NOE} + E_{HARM} + E_{CDIH} + E_{NCS} + E_{DG} + E_{RELA} + E_{PLAN}$$

$$E_{EMPIRICAL} = \sum_{p=1}^N [w_{BOND}^p E_{BOND} + w_{ANGL}^p E_{ANGL} + w_{DIHE}^p E_{DIHE} + w_{IMPR}^p E_{IMPR} + w_{VDW}^p E_{VDW} + w_{ELEC}^p E_{ELEC} + w_{PVDW}^p E_{PVDW} + w_{PELE}^p E_{PELE} + w_{HBON}^p E_{HBON}]$$

Difference Fourier

Obs. $\rho_o(x, y, z) = \frac{1}{V} \sum_h \sum_k \sum_l F_{o,hkl} e^{-2\pi i(hx+ky+lz)} + R$

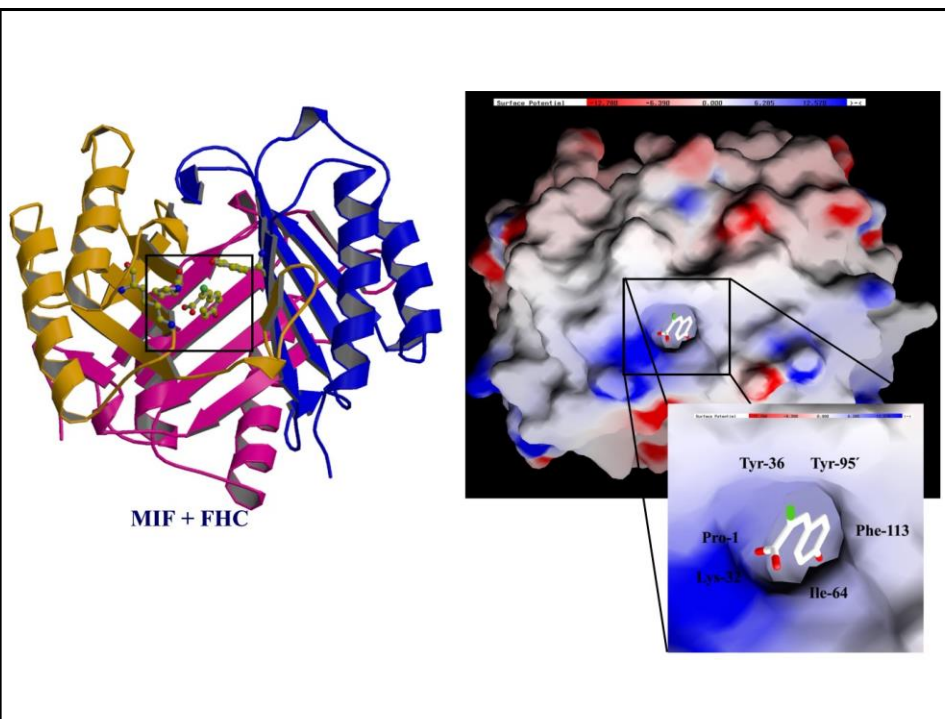
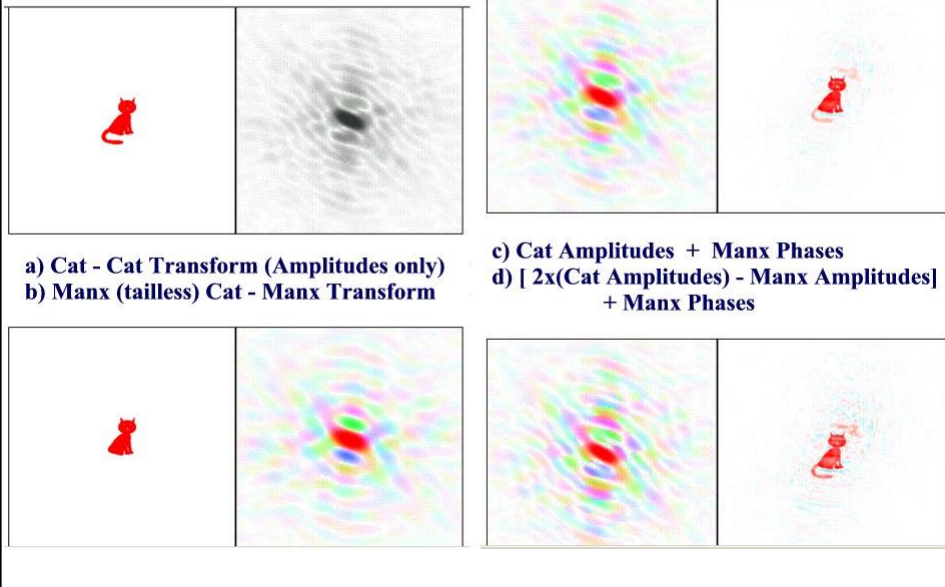
Calc. $\rho_c(x, y, z) = \frac{1}{V} \sum_h \sum_k \sum_l F_{c,hkl} e^{-2\pi i(hx+ky+lz)} + R'$

$$\rho_o(x, y, z) - \rho_c(x, y, z) = \frac{1}{V} \sum_h \sum_k \sum_l (F_o - F_c)_{hkl} e^{-2\pi i(hx+ky+lz)} + R - R'$$

$$\rho_o - \rho_c = \frac{1}{V} \sum_h \sum_k \sum_l \Delta F_{hkl} e^{-2\pi i(hx+ky+lz)}$$

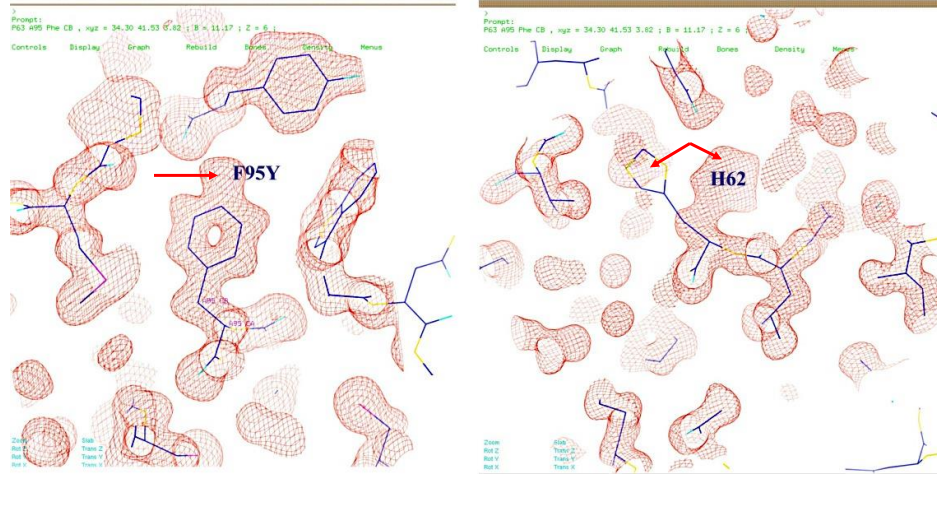
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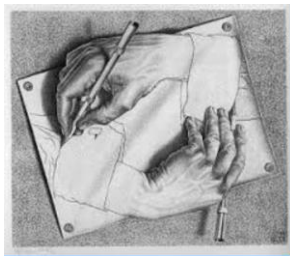
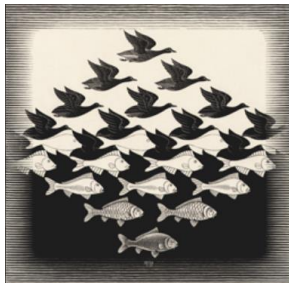


Examples of Difference Fouriers

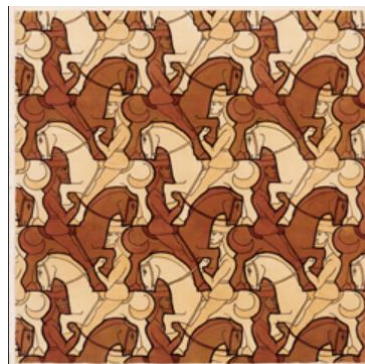
MIF - 1.5A 2Fo-Fc

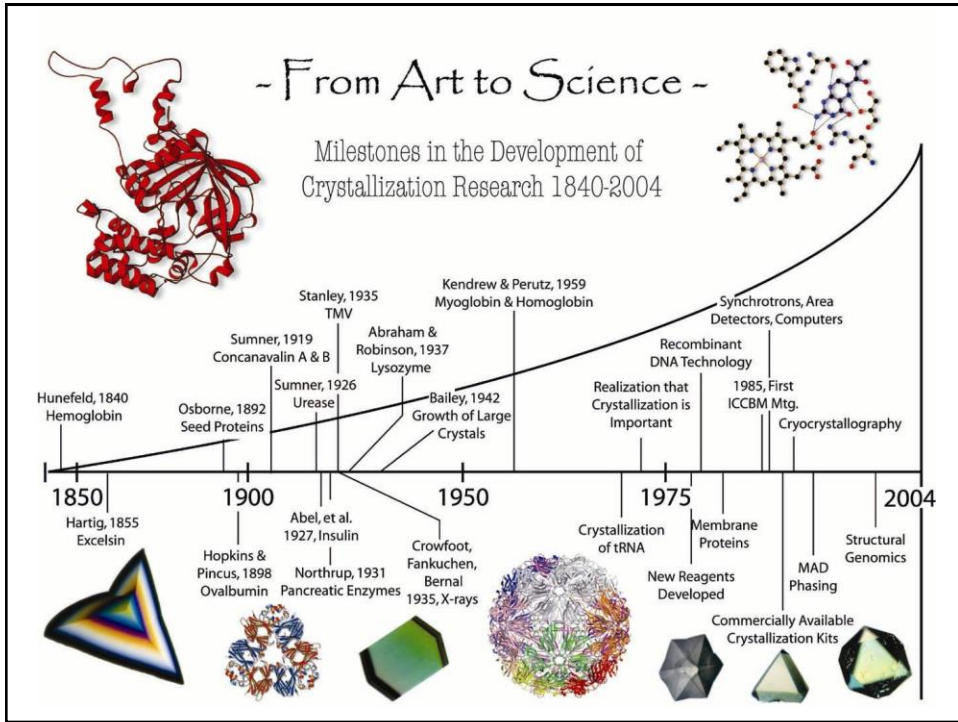


M.C. Escher



Escher sketch





2014 International Year of Crystallography

General Assembly

Recognizing that humankind's understanding of the material nature of our world is grounded, in particular, in our knowledge of crystallography.

Stressing that education about and the application of crystallography are critical in addressing challenges such as diseases and environmental problems, by providing protein and small molecule structures suited for drug design essential for medicine and public health, as well as solutions for plant and soil contamination.

Considering further that 2014 marks the centenary of the beginning of modern crystallography and its identification as the most powerful tool for structure determination of matter.

Being aware that 2014 provides an opportunity to promote international collaboration as part of the fifty-fifth anniversary of the founding of the International Union of Crystallography.

Noting the broader welcome by the crystallographic community worldwide of the idea of 2014 being designated as the International Year of Crystallography.

Recognizing the leading role of the International Union of Crystallography, an adhering body of the International Council for Science, in coordinating and promoting crystallographic activities at the international, regional and national levels around the world.

Decides to proclaim 2014 the International Year of Crystallography.

International Year of Crystallography 2014

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The International Year of Crystallography 2014 will be a year of celebration, not only for the centenary of the discovery of X-ray diffraction, which allowed the detailed analysis of crystalline matter, but also for the centenary of the discovery of the structure of DNA, which allowed the understanding of the genetic code. The International Year of Crystallography 2014 will be a year of celebration, not only for the centenary of the discovery of X-ray diffraction, which allowed the detailed analysis of crystalline matter, but also for the centenary of the discovery of the structure of DNA, which allowed the understanding of the genetic code.

The IUCr is working with UNESCO in leading efforts to celebrate the celebration of IYOC 2014. The major objectives of IYOC 2014 are to increase public awareness of the science of crystallography and how it underpins many technological developments in our modern society. IYOC 2014 will have a strong educational component aimed at students of all ages. The IUCr already supports workshops and lectures to provide training in crystallography, and during IYOC 2014 will organize schools and educational activities in all parts of the globe. In addition, IUCr-UNESCO open laboratories for students and young researchers are being planned in Africa, Asia and South America. The laboratories will comprise posters and lectures or workshops, along with hands-on experiments including the use of diffractometers. These activities are intended to increase global awareness of crystallography and, in the longer term, will have an impact on international collaborations and the worldwide development of science-based technologies.

See the IYOC website for more information: <http://www.iyoc2014.org>

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Nobel Laureates