

# Review Summary – CH370 / 387D - Exam 2

## Review of Nucleic Acids: Structures / Folding

Know N Bases; Primary & Secondary structure: double helix by Watson & Crick -1953  
Nucleotide pairings: Watson-Crick  
Conformations of nucleosides - syn / anti; Sugar pucker: endo or exo  
Stabilization (destabilization) Hydrogen Bonding / Electrostatics / Stacking  
Denatured DNA: Heat denaturation of DNA is called "melting,"  $T_m$  / hypochromism.

## Radioactivity and Counting

Radioactive decay processes ( $\alpha$  /  $\beta^+$  /  $\beta^-$  / E.C.); Radioactivity rays ( $\gamma$ -rays)  
Half life:  $A = A_0 \exp(-kt)$  where  $k = \ln 2 / \text{half-life}$   
Measurement of Radioactivity: Geiger Counter / Film / PhosphorImagers / LSC  
Liquid Scintillation Counting: Excited solvent /  $1^\circ$  "fluor" /  $2^\circ$  "fluor" / PM

## Electrophoresis – transport of charged particle in an electric field.

Theory:  $F_{\text{tot}} = qE - fv = ma = m(dv/dt) = 0$ ;  $v = (qE/f)$   
 $f = 6\pi\eta R$  for spheres;  $\eta = \text{Viscosity} \sim 0.01 \text{g}/(\text{cm}\cdot\text{sec})$   
Ferguson Plots: electrophoretic mobility reflects both charge and size/shape  
Methods: slab / tube / seq. gels / (native; denatured) / Disc. Gel / PAGE / PFGE / IEF / CE  
SDS-PAGE (subunit MW) / buffer system / stains; IEF gels / 2D-PAGE

## Centrifugation

Theory:  $F_{\text{tot}} = m_{\text{eff}}\omega^2 r - fv = m\omega^2 r(1 - v'\rho) - fv = ma = m(dv/dt) = 0$ ; ( $v'$  is "v bar")  
Preparative Methods: RCF / Rotors / Density Gradient: Zonal vs. Isopycnic Methods  
Analytical Methods / Modern **Analytical Ultracentrifuge**  
Optics: **Schlieren** ( $\alpha = aK(dc/dx)$ ); **Interference** ( $\Delta J = (aK\Delta c)/\lambda$ ); **Abspt. optics** ( $A \sim c$ )  
**Sedimentation Velocity**:  $s = v/\omega^2 r = (m(1 - v'\rho)/f)$ ;  $\rightarrow$  plot ( $\ln r$ ) vs.  $t \rightarrow$  slope =  $s\omega^2$   
Sed. Vel. plus Diffusion:  $D = (kT/f) = (RT/N^0 f)$ ;  $\rightarrow s = MD(1 - v'\rho)/RT$   
**Sedimentation Equil.**:  $\ln c_r - \ln c_m = [M\omega^2(1 - v'\rho)/(2RT)](r^2 - r_m^2) \rightarrow$  plot  $\ln c$  vs.  $r^2$

## CD

**Terms:** CD / ORD / Plane polarized light vs. Circularly polarized light. etc  
Special type of spectroscopy - meas. the difference in left and right handed absorbance:  
 $A(l) - A(r)$  or essentially looking at difference in  $\epsilon_L - \epsilon_R$   
The instrument: measurements in far UV **170-240 nm (proteins); 170-300 nm (nucleic acids)**.  
CD spectra can distinguish types of secondary structure (helix, sheet, r.coil / B-DNA, A-DNA) etc.  
Applications: Folding / Secondary Structure / Denaturation / Thermal Stability

## Light Scattering: "Static" and "Dynamic"

Rayleigh (Static) Scattering –  $i/I_0 = N[8\pi^4\alpha^2 / r^2\lambda^4](1 + \cos^2\theta)$  for unpolarized radiation.  
Raleigh Ratio:  $R_\theta = (i_\theta / I_0)(r^2 / (1 + \cos^2\theta)) = [2\pi^2 n_0^2 (dn/dc)^2 / \lambda^4 N_0^2] CM$  or  $R_\theta = KCM$   
 $KC/R_\theta = 1/(M^*P(\theta)) + 2 A_2C$ ; Mean Square Radius ( $R_g$ ) 10 nm to 150 nm  
Experimental (Use of LS and RI);  $LS = K_{LS}CM(dn/dc)^2$ ;  $RI = K_{RI}C(dn/dc)$   
or  $LS/RI = M[(K_{LS}/K_{RI})(dn/dc)]$  or  $M = K'(LS)/(RI)$   
Polydispersity ( $M_w/M_n$ ); If normalized,  $LS = RI$  for monomer but  $LS = 2*RI$  for dimer  
**Dynamic Light Scattering** –Hydrodynamic (Stokes) Radius ( $R_h$ ) 1.5 to 1000 nm  
Applications: MW, oligomerization, polydispersity / purity